

1,4-Dithiane

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| Other names: | 1,4-Dithiacyclohexane 1,4-Dithiin, tetrahydro- 1,4-dithian p-Dithiane p-Dithiin, tetrahydro- para-Dithiane |
| Inchi: | InChI=1S/C4H8S2/c1-2-6-4-3-5-1/h1-4H2 |
| InchiKey: | LOZWAPSEEHRYPG-UHFFFAOYSA-N |
| Formula: | C4H8S2 |
| SMILES: | C1CSCCS1 |
| Mol. weight [g/mol]: | 120.24 |
| CAS: | 505-29-3 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|-------------|--------|----------------|
| gf | 94.68 | kJ/mol | Joback Method |
| hf | 39.29 | kJ/mol | Joback Method |
| hfus | 4.19 | kJ/mol | Joback Method |
| hsub | 63.00 | kJ/mol | NIST Webbook |
| hsub | 68.90 | kJ/mol | NIST Webbook |
| hvap | 36.86 | kJ/mol | Joback Method |
| ie | 8.46 | eV | NIST Webbook |
| ie | 8.58 | eV | NIST Webbook |
| ie | 8.50 | eV | NIST Webbook |
| ie | 8.80 ± 0.10 | eV | NIST Webbook |
| log10ws | -1.16 | | Crippen Method |
| logp | 1.466 | | Crippen Method |
| mvol | 89.060 | ml/mol | McGowan Method |
| pc | 5281.57 | kPa | Joback Method |
| rinpol | 1019.00 | | NIST Webbook |
| rinpol | 1094.00 | | NIST Webbook |
| rinpol | 1067.70 | | NIST Webbook |
| rinpol | 1019.00 | | NIST Webbook |
| rinpol | 1060.20 | | NIST Webbook |
| rinpol | 1018.70 | | NIST Webbook |
| rinpol | 1060.20 | | NIST Webbook |
| rinpol | 1074.00 | | NIST Webbook |

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|--------|---------------|---------|---------------|
| rinpol | 1068.00 | | NIST Webbook |
| rinpol | 1019.00 | | NIST Webbook |
| rinpol | 1019.00 | | NIST Webbook |
| rinpol | 1074.00 | | NIST Webbook |
| rinpol | 1060.20 | | NIST Webbook |
| rinpol | 1038.00 | | NIST Webbook |
| rinpol | 1050.00 | | NIST Webbook |
| rinpol | 1038.00 | | NIST Webbook |
| rinpol | 1028.00 | | NIST Webbook |
| rinpol | 1022.00 | | NIST Webbook |
| ripol | 1641.00 | | NIST Webbook |
| ripol | 1618.90 | | NIST Webbook |
| ripol | 1575.70 | | NIST Webbook |
| ripol | 1576.00 | | NIST Webbook |
| ripol | 1618.00 | | NIST Webbook |
| ripol | 1576.00 | | NIST Webbook |
| ripol | 1618.00 | | NIST Webbook |
| ripol | 1575.70 | | NIST Webbook |
| tb | 472.70 | K | NIST Webbook |
| tc | 657.97 | K | Joback Method |
| tf | 384.60 ± 0.20 | K | NIST Webbook |
| vc | 0.285 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 190.10 | J/molxK | 575.58 | Joback Method |
| cpg | 180.29 | J/molxK | 534.38 | Joback Method |
| cpg | 169.74 | J/molxK | 493.19 | Joback Method |
| cpg | 158.39 | J/molxK | 451.99 | Joback Method |
| cpg | 207.62 | J/molxK | 657.97 | Joback Method |
| cpg | 199.20 | J/molxK | 616.77 | Joback Method |
| cpg | 146.23 | J/molxK | 410.80 | Joback Method |
| cps | 129.80 | J/molxK | 300.00 | NIST Webbook |
| hfust | 21.60 | kJ/mol | 384.60 | NIST Webbook |
| hfust | 21.60 | kJ/mol | 384.60 | NIST Webbook |
| hfust | 21.60 | kJ/mol | 384.60 | NIST Webbook |
| hsubt | 72.40 | kJ/mol | 264.50 | NIST Webbook |
| hvapt | 47.90 | kJ/mol | 412.50 | NIST Webbook |
| hvapt | 48.70 | kJ/mol | 413.00 | NIST Webbook |

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|-------|-------|---------|--------|--------------------------------------|
| psub | 4.63 | kPa | 376.95 | Vapor Pressure of Solid 1,4-Dithiane |
| psub | 3.91 | kPa | 374.25 | Vapor Pressure of Solid 1,4-Dithiane |
| psub | 2.65 | kPa | 367.35 | Vapor Pressure of Solid 1,4-Dithiane |
| psub | 1.89 | kPa | 362.35 | Vapor Pressure of Solid 1,4-Dithiane |
| psub | 1.33 | kPa | 359.45 | Vapor Pressure of Solid 1,4-Dithiane |
| psub | 1.07 | kPa | 355.75 | Vapor Pressure of Solid 1,4-Dithiane |
| psub | 3.83 | kPa | 374.05 | Vapor Pressure of Solid 1,4-Dithiane |
| psub | 3.25 | kPa | 370.65 | Vapor Pressure of Solid 1,4-Dithiane |
| sfust | 56.20 | J/molxK | 384.60 | NIST Webbook |

Correlations

| Information | Value |
|-----------------------------|-------------------------------|
| Property code | pvap |
| Equation | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A | 1.01308e+01 |
| Coeff. B | -1.50998e+03 |
| Coeff. C | -1.98778e+02 |
| Temperature range (K), min. | 352.18 |
| Temperature range (K), max. | 512.10 |

Sources

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|--|---|
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Vapor Pressure of Solid 1,4-Dithiane: | https://www.doi.org/10.1021/je800586u |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C505293&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| cps: | Solid phase heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hfust: | Enthalpy of fusion at a given temperature |
| hsub: | Enthalpy of sublimation at standard conditions |
| hsubt: | Enthalpy of sublimation at a given temperature |
| hvap: | Enthalpy of vaporization at standard conditions |
| hvapt: | Enthalpy of vaporization at a given temperature |
| ie: | Ionization energy |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| psub: | Sublimation pressure |
| pvap: | Vapor pressure |
| rinpol: | Non-polar retention indices |
| ripol: | Polar retention indices |
| sfust: | Entropy of fusion at a given temperature |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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