

# 2-Coumaranone

<b>Other names:</b>	(3H)-benzofuran-2-one .alpha.-coumaranone 2(3H)-Benzofuranone 2-Coumaronone 2-benzofuranone 2-coumarone 3H-benzofuran-2-one Benzofuranone benzo[b]furan-2(3H)-one
<b>Inchi:</b>	InChI=1S/C8H6O2/c9-8-5-6-3-1-2-4-7(6)10-8/h1-4H,5H2
<b>InchiKey:</b>	ACZGCWSMSTYWDQ-UHFFFAOYSA-N
<b>Formula:</b>	C8H6O2
<b>SMILES:</b>	O=C1Cc2ccccc2O1
<b>Mol. weight [g/mol]:</b>	134.13
<b>CAS:</b>	553-86-6

## Physical Properties

Property code	Value	Unit	Source
gf	-20.99	kJ/mol	Joback Method
hf	-159.95	kJ/mol	Joback Method
hfus	14.68	kJ/mol	Joback Method
hvap	45.32	kJ/mol	Joback Method
log10ws	-1.65		Crippen Method
logp	1.148		Crippen Method
mcvol	96.400	ml/mol	McGowan Method
pc	4559.21	kPa	Joback Method
rinpol	1250.00		NIST Webbook
rinpol	1289.00		NIST Webbook
ripol	2352.00		NIST Webbook
tb	522.20	K	NIST Webbook
tb	520.00 ± 2.00	K	NIST Webbook
tc	767.66	K	Joback Method
tf	301.60	K	NIST Webbook
vc	0.361	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	259.45	J/mol×K	726.43	Joback Method
cpg	208.42	J/mol×K	520.28	Joback Method
cpg	220.19	J/mol×K	561.51	Joback Method
cpg	231.14	J/mol×K	602.74	Joback Method
cpg	241.30	J/mol×K	643.97	Joback Method
cpg	250.73	J/mol×K	685.20	Joback Method
cpg	267.50	J/mol×K	767.66	Joback Method
hvapt	78.70	kJ/mol	298.15	Energetics of 2- and 3-coumaranone isomers: A combined calorimetric and computational study

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	326.00	K	0.01	NIST Webbook

# Sources

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C553866&Units=SI>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method: [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Energetics of 2- and 3-coumaranone isomers: A combined calorimetric and computational study: <https://www.doi.org/10.1016/j.jct.2013.08.012>

Joback Method: [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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