

# 2,6-Dichlorobenzyl alcohol, 2-methylbutyl ether

**Inchi:** InChI=1S/C12H16Cl2O/c1-3-9(2)7-15-8-10-11(13)5-4-6-12(10)14/h4-6,9H,3,7-8H2,1-2H3  
**InchiKey:** DZWFKBVEZWPWPY-UHFFFAOYSA-N  
**Formula:** C12H16Cl2O  
**SMILES:** CCC(C)COCc1c(Cl)cccc1Cl  
**Mol. weight [g/mol]:** 247.16

## Physical Properties

Property code	Value	Unit	Source
gf	12.01	kJ/mol	Joback Method
hf	-246.40	kJ/mol	Joback Method
hfus	26.16	kJ/mol	Joback Method
hvap	56.70	kJ/mol	Joback Method
log10ws	-4.66		Crippen Method
logp	4.556		Crippen Method
mcvol	186.530	ml/mol	McGowan Method
pc	2159.31	kPa	Joback Method
rinpol	1633.00		NIST Webbook
rinpol	1633.00		NIST Webbook
tb	607.44	K	Joback Method
tc	820.11	K	Joback Method
tf	343.53	K	Joback Method
vc	0.710	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	430.61	J/molxK	607.44	Joback Method
cpg	445.15	J/molxK	642.88	Joback Method
cpg	458.88	J/molxK	678.33	Joback Method
cpg	471.81	J/molxK	713.77	Joback Method
cpg	483.97	J/molxK	749.22	Joback Method
cpg	495.37	J/molxK	784.66	Joback Method
cpg	506.02	J/molxK	820.11	Joback Method
dvisc	0.0015502	Paxs	343.53	Joback Method

dvisc	0.0008347	Paxs	387.51	Joback Method
dvisc	0.0005098	Paxs	431.50	Joback Method
dvisc	0.0003412	Paxs	475.49	Joback Method
dvisc	0.0002444	Paxs	519.47	Joback Method
dvisc	0.0001844	Paxs	563.46	Joback Method
dvisc	0.0001449	Paxs	607.44	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378132&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378132&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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