

# Methyl (Z)-dihydrojasmonate

<b>Other names:</b>	cis-Methyl dihydrojasmonate
<b>Inchi:</b>	InChI=1S/C13H22O3/c1-3-4-5-6-11-10(7-8-12(11)14)9-13(15)16-2/h10-11H,3-9H2,1-2H3
<b>InchiKey:</b>	KVWWIYGFBYDJQC-QWRGUYRKSA-N
<b>Formula:</b>	C13H22O3
<b>SMILES:</b>	CCCCC1C(=O)CCC1CC(=O)OC
<b>Mol. weight [g/mol]:</b>	226.31

## Physical Properties

Property code	Value	Unit	Source
gf	-269.09	kJ/mol	Joback Method
hf	-654.01	kJ/mol	Joback Method
hfus	26.73	kJ/mol	Joback Method
hvap	57.88	kJ/mol	Joback Method
log10ws	-2.82		Crippen Method
logp	2.725		Crippen Method
mcvol	192.180	ml/mol	McGowan Method
pc	1996.55	kPa	Joback Method
rinpol	1656.00		NIST Webbook
rinpol	1656.00		NIST Webbook
rinpol	1654.00		NIST Webbook
rinpol	1656.00		NIST Webbook
ripol	2287.00		NIST Webbook
ripol	2287.00		NIST Webbook
tb	651.56	K	Joback Method
tc	854.07	K	Joback Method
tf	383.31	K	Joback Method
vc	0.735	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	544.32	J/molxK	651.56	Joback Method
cpg	562.70	J/molxK	685.31	Joback Method
cpg	580.11	J/molxK	719.06	Joback Method

cpg	596.55	J/mol×K	752.81	Joback Method
cpg	612.00	J/mol×K	786.57	Joback Method
cpg	626.46	J/mol×K	820.32	Joback Method
cpg	639.93	J/mol×K	854.07	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R417746&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R417746&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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