

Benzene, 1,3-butadienyl-

Other names:	1,3-Butadiene, 1-phenyl- 1-Phenylbutadiene 1-Phenyl-1,3-butadiene 1,3-Butadienylbenzene
Inchi:	InChI=1S/C10H10/c1-2-3-7-10-8-5-4-6-9-10/h2-9H,1H2
InchiKey:	XZKRX PZXQLARHH-UHFFFAOYSA-N
Formula:	C10H10
SMILES:	<chem>C=CC=Cc1ccccc1</chem>
Mol. weight [g/mol]:	130.19
CAS:	1515-78-2

Physical Properties

Property code	Value	Unit	Source
gf	313.79	kJ/mol	Joback Method
hf	229.45	kJ/mol	Joback Method
hfus	14.62	kJ/mol	Joback Method
hvap	39.42	kJ/mol	Joback Method
ie	8.39	eV	NIST Webbook
log10ws	-2.99		Crippen Method
logp	2.886		Crippen Method
mcvol	119.400	ml/mol	McGowan Method
pc	3254.14	kPa	Joback Method
rinpol	177.00		NIST Webbook
rinpol	1150.00		NIST Webbook
ripol	1572.00		NIST Webbook
tb	455.72	K	Joback Method
tc	675.61	K	Joback Method
tf	222.04	K	Joback Method
vc	0.449	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	223.50	J/molxK	455.72	Joback Method

cpg	237.60	J/mol×K	492.37	Joback Method
cpg	250.71	J/mol×K	529.02	Joback Method
cpg	262.90	J/mol×K	565.67	Joback Method
cpg	274.22	J/mol×K	602.32	Joback Method
cpg	284.74	J/mol×K	638.97	Joback Method
cpg	294.51	J/mol×K	675.61	Joback Method
dvisc	0.0030060	Paxs	222.04	Joback Method
dvisc	0.0013450	Paxs	260.99	Joback Method
dvisc	0.0007416	Paxs	299.93	Joback Method
dvisc	0.0004689	Paxs	338.88	Joback Method
dvisc	0.0003258	Paxs	377.83	Joback Method
dvisc	0.0002424	Paxs	416.77	Joback Method
dvisc	0.0001896	Paxs	455.72	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C1515782&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature

tc: Critical Temperature
tf: Normal melting (fusion) point
vc: Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/67-775-8/Benzene-1-3-butadienyl.pdf>

Generated by Cheméo on 2024-04-24 02:32:27.146900067 +0000 UTC m=+16215196.067477379.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.