

# Methyldinitramine

Inchi:	InChI=1S/CH3N3O4/c1-2(3(5)6)4(7)8/h1H3
InchiKey:	AAIWMHUYNFKAM-UHFFFAOYSA-N
Formula:	CH3N3O4
SMILES:	CN([N+](=O)[O-])[N+](=O)[O-]
Mol. weight [g/mol]:	121.05
CAS:	25346-05-8

## Physical Properties

Property code	Value	Unit	Source
chs	-823.80 ± 0.70	kJ/mol	NIST Webbook
gf	139.42	kJ/mol	Joback Method
hf	53.50 ± 0.80	kJ/mol	NIST Webbook
hf	43.10 ± 6.30	kJ/mol	NIST Webbook
hfl	1.70 ± 0.80	kJ/mol	NIST Webbook
hfs	1.50 ± 0.70	kJ/mol	NIST Webbook
hfus	24.09	kJ/mol	Joback Method
hsub	52.00 ± 0.40	kJ/mol	NIST Webbook
hvap	41.40 ± 5.40	kJ/mol	NIST Webbook
log10ws	-0.96		Crippen Method
logp	-0.698		Crippen Method
mcvol	69.770	ml/mol	McGowan Method
pc	5990.66	kPa	Joback Method
tb	538.40	K	Joback Method
tc	786.86	K	Joback Method
tf	420.72	K	Joback Method
vc	0.274	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	154.34	J/mol×K	538.40	Joback Method
cpg	160.92	J/mol×K	579.81	Joback Method
cpg	166.93	J/mol×K	621.22	Joback Method
cpg	172.41	J/mol×K	662.63	Joback Method

cpg	177.39	J/mol×K	704.04	Joback Method
cpg	181.91	J/mol×K	745.45	Joback Method
cpg	185.99	J/mol×K	786.86	Joback Method

## Sources

Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C25346058&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C25346058&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/67-788-4/Methyldinitramine.pdf>

Generated by Cheméo on 2024-04-28 14:38:09.061820964 +0000 UTC m=+16604337.982398277.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.