

3-Methyl-3-sulfanylbutyl Tetradecanoate

Inchi: InChI=1S/C19H38O2S/c1-4-5-6-7-8-9-10-11-12-13-14-15-18(20)21-17-16-19(2,3)22/h22
InchiKey: FXRSTAIWMGQKTN-UHFFFAOYSA-N
Formula: C19H38O2S
SMILES: CCCCCCCCCCCCCC(=O)OCCC(C)(C)S
Mol. weight [g/mol]: 330.57

Physical Properties

Property code	Value	Unit	Source
gf	-92.59	kJ/mol	Joback Method
hf	-650.56	kJ/mol	Joback Method
hfus	44.38	kJ/mol	Joback Method
hvap	72.48	kJ/mol	Joback Method
log10ws	-6.82		Crippen Method
logp	6.329		Crippen Method
mcvol	302.360	ml/mol	McGowan Method
pc	1178.47	kPa	Joback Method
rinpol	2287.00		NIST Webbook
rinpol	2287.00		NIST Webbook
ripol	2770.00		NIST Webbook
ripol	2770.00		NIST Webbook
tb	770.04	K	Joback Method
tc	957.18	K	Joback Method
tf	414.93	K	Joback Method
vc	1.167	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	916.80	J/molxK	770.04	Joback Method
cpg	935.70	J/molxK	801.23	Joback Method
cpg	953.58	J/molxK	832.42	Joback Method
cpg	970.48	J/molxK	863.61	Joback Method
cpg	986.44	J/molxK	894.80	Joback Method
cpg	1001.51	J/molxK	925.99	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R519629&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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