

Butane, 1,3-dichloro-3-methyl-

Other names:	1,3-Dichloro-3-methylbutane
Inchi:	InChI=1S/C5H10Cl2/c1-5(2,7)3-4-6/h3-4H2,1-2H3
InchiKey:	QGCOTJZDMVMLRQ-UHFFFAOYSA-N
Formula:	C5H10Cl2
SMILES:	CC(C)(Cl)CCCl
Mol. weight [g/mol]:	141.04
CAS:	624-96-4

Physical Properties

Property code	Value	Unit	Source
gf	-29.80	kJ/mol	Joback Method
hf	-186.76	kJ/mol	Joback Method
hfus	9.69	kJ/mol	Joback Method
hvap	34.20	kJ/mol	Joback Method
log10ws	-2.33		Crippen Method
logp	2.633		Crippen Method
mcvol	105.790	ml/mol	McGowan Method
pc	3254.14	kPa	Joback Method
tb	385.43	K	Joback Method
tc	579.04	K	Joback Method
tf	208.37	K	Joback Method
vc	0.403	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	177.40	J/mol×K	385.43	Joback Method
cpg	222.52	J/mol×K	546.77	Joback Method
cpg	214.58	J/mol×K	514.50	Joback Method
cpg	206.13	J/mol×K	482.23	Joback Method
cpg	197.13	J/mol×K	449.97	Joback Method
cpg	187.56	J/mol×K	417.70	Joback Method
cpg	229.97	J/mol×K	579.04	Joback Method
dvisc	0.0003821	Pa×s	385.43	Joback Method

dvisc	0.0005124	Pa×s	355.92	Joback Method
dvisc	0.0007246	Pa×s	326.41	Joback Method
dvisc	0.0010979	Pa×s	296.90	Joback Method
dvisc	0.0018231	Pa×s	267.39	Joback Method
dvisc	0.0034334	Pa×s	237.88	Joback Method
dvisc	0.0077359	Pa×s	208.37	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.32583e+01
Coeff. B	-2.82647e+03
Coeff. C	-9.56000e+01
Temperature range (K), min.	313.51
Temperature range (K), max.	451.27

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C624964&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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