

# Verapamil

**Other names:**

5-((3,4-Dimethoxyphenethyl)methylamino)-2-(3,4-dimethoxyphenyl)-2-isopropylvaleronitril  
Benzeneacetonitrile,  
«alpha»-[3-[[2-(3,4-dimethoxyphenyl)ethyl]methylamino]propyl]-3,4-dimethoxy-«alpha»-(  
Benzeneacetonitrile,  
«alpha»-[3-[[2-(3,4-dimethoxyphenyl)ethyl]methylamino]propyl]-3,4-dimethoxy-«alpha»-  
CP-16533-1  
D-365  
Dilacorán  
Iproveratríl  
Valeronitril,  
5-((3,4-dimethoxyphenethyl)methylamino)-2-(3,4-dimethoxyphenyl)-2-isopropyl-  
Vasolan

**Inchi:** InChI=1S/C27H38N2O4/c1-20(2)27(19-28,22-10-12-24(31-5)26(18-22)33-7)14-8-15-29(3  
**InchiKey:** SGTNSNPWRIOYBX-UHFFFAOYSA-N  
**Formula:** C27H38N2O4  
**SMILES:** COc1ccc(CCN(C)CCCC(C#N)(c2ccc(OC)c(OC)c2)C(C)C)cc1OC  
**Mol. weight [g/mol]:** 454.60  
**CAS:** 52-53-9

## Physical Properties

Property code	Value	Unit	Source
gf	187.12	kJ/mol	Joback Method
hf	-483.93	kJ/mol	Joback Method
hfus	50.55	kJ/mol	Joback Method
hvap	103.37	kJ/mol	Joback Method
log10ws	-4.15		Aqueous Solubility Prediction Method
logp	5.093		Crippen Method
mcvol	378.610	ml/mol	McGowan Method
pc	931.78	kPa	Joback Method
rinpol	3161.00		NIST Webbook
rinpol	3150.00		NIST Webbook
rinpol	3150.00		NIST Webbook
tb	1090.97	K	Joback Method
tc	1335.66	K	Joback Method
tf	469.48	K	Aqueous Solubility Prediction Method
vc	1.431	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1288.74	J/mol×K	1090.97	Joback Method
cpg	1301.36	J/mol×K	1131.75	Joback Method
cpg	1312.12	J/mol×K	1172.53	Joback Method
cpg	1321.07	J/mol×K	1213.32	Joback Method
cpg	1328.27	J/mol×K	1254.10	Joback Method
cpg	1333.79	J/mol×K	1294.88	Joback Method
cpg	1337.68	J/mol×K	1335.66	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C52539&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C52539&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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