

# Bicyclo[2.2.2]-7-octene-2,3,5,6-tetracarboxylic acid dianhydride

<b>Other names:</b>	4,8-Etheno-1H,3H-benzo[1,2-c:4,5-c']difuran-1,3,5,7-tetrone, 3a,4,4a,7a,8,8a-hexahydro- bicyclo[2.2.2]oct-7-ene-2,3,5,6-tetracarboxylic dianhydride
<b>Inchi:</b>	InChI=1S/C12H8O6/c13-9-5-3-1-2-4(7(5)11(15)17-9)8-6(3)10(14)18-12(8)16/h1-8H
<b>InchiKey:</b>	XLOGCGOPKPCECW-UHFFFAOYSA-N
<b>Formula:</b>	C12H8O6
<b>SMILES:</b>	O=C1OC(=O)C2C3C=CC(C12)C1C(=O)OC(=O)C31
<b>Mol. weight [g/mol]:</b>	248.19
<b>CAS:</b>	1719-83-1

## Physical Properties

Property code	Value	Unit	Source
gf	-379.10	kJ/mol	Joback Method
hf	-809.83	kJ/mol	Joback Method
hfus	32.54	kJ/mol	Joback Method
hvap	67.98	kJ/mol	Joback Method
log10ws	-0.10		Crippen Method
logp	-0.566		Crippen Method
mvol	150.220	ml/mol	McGowan Method
pc	3427.87	kPa	Joback Method
tb	824.46	K	Joback Method
tc	1100.22	K	Joback Method
tf	608.02	K	Joback Method
vc	0.574	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	529.40	J/molxK	824.46	Joback Method
cpg	544.83	J/molxK	870.42	Joback Method
cpg	558.45	J/molxK	916.38	Joback Method
cpg	570.24	J/molxK	962.34	Joback Method
cpg	580.16	J/molxK	1008.30	Joback Method
cpg	588.19	J/molxK	1054.26	Joback Method
cpg	594.29	J/molxK	1100.22	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1719831&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1719831&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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