

# Cyacetacide

## Other names:

Cyanoacetohydrazide  
2-Cyanoacetylhydrazide  
Cyanoacethydrazide  
Cyanoacetic acid hydrazide  
Acetic acid, cyano-, hydrazide  
«alpha»-Cyanoacetohydrazide  
(Cyanoacetyl)hydrazine  
Armazal  
AB-42  
Cianazil  
Cyacetacid  
Cyacetazid  
Cyacetazide  
Cyanacethydrazide  
Cyanacetic acid, hydrazide  
Cyanacetohydrazide  
Cyanacetylhydrazide  
Cyanazide  
Cyanizide  
Cyanoacetylhydrazide  
Cyanoethydrazide  
Cyazid  
Cyazide  
Dictycide  
Dictyzide  
Helmox  
Hidacian  
Hidaciann  
Leandin  
Mackreazid  
Malonitrile hydrazide  
Malononitrile hydrazide  
Neohydrazid  
Reacid  
Reazid  
Reazide  
Tsiazid  
USAF KF-18  
Kyanacethydrazid  
(«alpha»-Cyanoacetyl)hydrazine

2-Cyanoacetylhydrazide  
2-Cyanoacetohydrazide  
Alpha-cyanoacetic acid, hydrazide  
Cyanoacetic hydrazide

**Inchi:** InChI=1S/C3H5N3O/c4-2-1-3(7)6-5/h1,5H2,(H,6,7)  
**InchiKey:** HPHBOJANXDKUQD-UHFFFAOYSA-N  
**Formula:** C3H5N3O  
**SMILES:** N#CCC(=O)NN  
**Mol. weight [g/mol]:** 99.09  
**CAS:** 140-87-4

## Physical Properties

Property code	Value	Unit	Source
gf	134.48	kJ/mol	Joback Method
hf	34.31	kJ/mol	Joback Method
hfus	16.93	kJ/mol	Joback Method
hvap	56.57	kJ/mol	Joback Method
log10ws	-0.33		Crippen Method
logp	-1.110		Crippen Method
mcvol	76.040	ml/mol	McGowan Method
pc	5116.65	kPa	Joback Method
tb	546.69	K	Joback Method
tc	766.75	K	Joback Method
tf	374.41	K	Joback Method
vc	0.299	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	162.94	J/molxK	546.69	Joback Method
cpg	168.72	J/molxK	583.37	Joback Method
cpg	174.16	J/molxK	620.04	Joback Method
cpg	179.27	J/molxK	656.72	Joback Method
cpg	184.05	J/molxK	693.40	Joback Method
cpg	188.51	J/molxK	730.08	Joback Method
cpg	192.66	J/molxK	766.75	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C140874&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C140874&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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