

# 4-Methoxybenzene-1,3-diamine, N1,N1,N3-triacetyl-

<b>Inchi:</b>	InChI=1S/C13H16N2O4/c1-8(16)14-12-7-11(5-6-13(12)19-4)15(9(2)17)10(3)18/h5-7H,1-
<b>InchiKey:</b>	DQPHCEHTPHJVYPY-UHFFFAOYSA-N
<b>Formula:</b>	C13H16N2O4
<b>SMILES:</b>	COc1ccc(N(C(C)=O)C(C)=O)cc1NC(C)=O
<b>Mol. weight [g/mol]:</b>	264.28

## Physical Properties

Property code	Value	Unit	Source
gf	-139.86	kJ/mol	Joback Method
hf	-447.02	kJ/mol	Joback Method
hfus	36.79	kJ/mol	Joback Method
hvap	79.26	kJ/mol	Joback Method
log10ws	-2.09		Crippen Method
logp	1.553		Crippen Method
mcvol	200.810	ml/mol	McGowan Method
pc	2545.61	kPa	Joback Method
rinsol	2258.00		NIST Webbook
tb	780.12	K	Joback Method
tc	995.18	K	Joback Method
tf	544.88	K	Joback Method
vc	0.745	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	565.53	J/mol×K	780.12	Joback Method
cpg	577.70	J/mol×K	815.96	Joback Method
cpg	588.93	J/mol×K	851.81	Joback Method
cpg	599.24	J/mol×K	887.65	Joback Method
cpg	608.66	J/mol×K	923.49	Joback Method
cpg	617.21	J/mol×K	959.33	Joback Method
cpg	624.91	J/mol×K	995.18	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373209&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373209&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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