

# Benzene, 1-(chloromethyl)-4-(1,1-dimethylethyl)-

Other names:	p-tert-Butyl-«alpha»-chlorotoluene p-tert-Butylbenzyl chloride 4-(tert-Butyl)benzyl chloride Benzene,1-(chloromethyl)-4-dimethylethyl)- p-(tert-butyl)-alpha-chlorotoluene
Inchi:	InChI=1S/C11H15Cl/c1-11(2,3)10-6-4-9(8-12)5-7-10/h4-7H,8H2,1-3H3
InchiKey:	WAXIFMGAKWIFDQ-UHFFFAOYSA-N
Formula:	C11H15Cl
SMILES:	CC(C)(C)c1ccc(CCl)cc1
Mol. weight [g/mol]:	182.69
CAS:	19692-45-6

## Physical Properties

Property code	Value	Unit	Source
gf	135.43	kJ/mol	Joback Method
hf	-69.80	kJ/mol	Joback Method
hfus	14.68	kJ/mol	Joback Method
hvap	46.11	kJ/mol	Joback Method
ie	8.60 ± 0.03	eV	NIST Webbook
log10ws	-3.82		Crippen Method
logp	3.723		Crippen Method
mcvol	154.330	ml/mol	McGowan Method
pc	2561.10	kPa	Joback Method
rinpol	1326.00		NIST Webbook
tb	516.94	K	Joback Method
tc	739.16	K	Joback Method
tf	285.01	K	Joback Method
vc	0.582	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	333.36	J/mol×K	516.94	Joback Method
cpg	349.41	J/mol×K	553.98	Joback Method

cpg	364.38	J/molxK	591.01	Joback Method
cpg	378.31	J/molxK	628.05	Joback Method
cpg	391.28	J/molxK	665.09	Joback Method
cpg	403.35	J/molxK	702.12	Joback Method
cpg	414.58	J/molxK	739.16	Joback Method
dvisc	0.0032297	Paxs	285.01	Joback Method
dvisc	0.0015576	Paxs	323.67	Joback Method
dvisc	0.0008777	Paxs	362.32	Joback Method
dvisc	0.0005524	Paxs	400.98	Joback Method
dvisc	0.0003772	Paxs	439.63	Joback Method
dvisc	0.0002739	Paxs	478.29	Joback Method
dvisc	0.0002087	Paxs	516.94	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	376.20	K	0.90	NIST Webbook

## Sources

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C19692456&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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