

Ethane, 2-bromo-1,1-dimethoxy-

Other names:	Acetaldehyde, bromo-, dimethyl acetal Bromoacetal Bromoacetaldehyde dimethyl acetal 2,2-Dimethoxybromoethane Dimethylbromoacetal 1-Bromo-2,2-dimethoxyethane 2-Bromo-1,1-dimethoxyethane 2-Bromoacetaldehyde dimethyl acetal 2,2-Dimethoxyethyl bromide
Inchi:	InChI=1S/C4H9BrO2/c1-6-4(3-5)7-2/h4H,3H2,1-2H3
InchiKey:	FUSFWUFSEJXMRQ-UHFFFAOYSA-N
Formula:	C4H9BrO2
SMILES:	COC(CBr)OC
Mol. weight [g/mol]:	169.02
CAS:	7252-83-7

Physical Properties

Property code	Value	Unit	Source
gf	-215.32	kJ/mol	Joback Method
hf	-369.28	kJ/mol	Joback Method
hfus	10.25	kJ/mol	Joback Method
hvap	35.37	kJ/mol	Joback Method
log10ws	-0.71		Crippen Method
logp	1.000		Crippen Method
mcvol	96.460	ml/mol	McGowan Method
pc	4114.41	kPa	Joback Method
tb	422.20	K	NIST Webbook
tc	591.85	K	Joback Method
tf	224.10	K	Joback Method
vc	0.351	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	167.78	J/molxK	401.48	Joback Method
cpg	175.46	J/molxK	433.21	Joback Method
cpg	182.95	J/molxK	464.94	Joback Method
cpg	190.23	J/molxK	496.66	Joback Method
cpg	197.30	J/molxK	528.39	Joback Method
cpg	204.14	J/molxK	560.12	Joback Method
cpg	210.76	J/molxK	591.85	Joback Method
dvisc	0.0032247	Paxs	224.10	Joback Method
dvisc	0.0016800	Paxs	253.66	Joback Method
dvisc	0.0010028	Paxs	283.23	Joback Method
dvisc	0.0006599	Paxs	312.79	Joback Method
dvisc	0.0004668	Paxs	342.35	Joback Method
dvisc	0.0003489	Paxs	371.92	Joback Method
dvisc	0.0002722	Paxs	401.48	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C7252837&Units=SI

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point

vc: Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/68-293-2/Ethane-2-bromo-1-1-dimethoxy.pdf>

Generated by Cheméo on 2024-04-27 16:03:52.505990162 +0000 UTC m=+16523081.426567479.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.