

# Fumaric acid, isobutyl pent-4-en-2-yl ester

<b>Inchi:</b>	InChI=1S/C13H20O4/c1-5-6-11(4)17-13(15)8-7-12(14)16-9-10(2)3/h5,7-8,10-11H,1,6,9H
<b>InchiKey:</b>	WXNYTZSPSXNNFE-BQYQJAHWSA-N
<b>Formula:</b>	C13H20O4
<b>SMILES:</b>	C=CCC(C)OC(=O)C=CC(=O)OCC(C)C
<b>Mol. weight [g/mol]:</b>	240.30

## Physical Properties

Property code	Value	Unit	Source
gf	-246.08	kJ/mol	Joback Method
hf	-569.16	kJ/mol	Joback Method
hfus	26.88	kJ/mol	Joback Method
hvap	61.36	kJ/mol	Joback Method
log10ws	-2.57		Crippen Method
logp	2.250		Crippen Method
mcvol	200.310	ml/mol	McGowan Method
pc	1950.95	kPa	Joback Method
rinsol	1552.00		NIST Webbook
tb	649.38	K	Joback Method
tc	840.40	K	Joback Method
tf	343.75	K	Joback Method
vc	0.760	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	525.27	J/molxK	649.38	Joback Method
cpg	539.89	J/molxK	681.22	Joback Method
cpg	553.76	J/molxK	713.05	Joback Method
cpg	566.88	J/molxK	744.89	Joback Method
cpg	579.27	J/molxK	776.72	Joback Method
cpg	590.95	J/molxK	808.56	Joback Method
cpg	601.94	J/molxK	840.40	Joback Method
dvisc	0.0023112	Paxs	343.75	Joback Method
dvisc	0.0009902	Paxs	394.69	Joback Method

dvisc	0.0005149	Paxs	445.63	Joback Method
dvisc	0.0003062	Paxs	496.56	Joback Method
dvisc	0.0002006	Paxs	547.50	Joback Method
dvisc	0.0001412	Paxs	598.44	Joback Method
dvisc	0.0001050	Paxs	649.38	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348922&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348922&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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