

# o-Anisic acid, heptadecyl ester

**Inchi:** InChI=1S/C25H42O3/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-19-22-28-25(26)23-20-17-  
**InchiKey:** CDHBRALYVRZRGF-UHFFFAOYSA-N  
**Formula:** C25H42O3  
**SMILES:** CCCCCCCCCCCCCCCCCOC(=O)c1ccccc1OC  
**Mol. weight [g/mol]:** 390.60

## Physical Properties

Property code	Value	Unit	Source
gf	-76.52	kJ/mol	Joback Method
hf	-711.29	kJ/mol	Joback Method
hfus	58.13	kJ/mol	Joback Method
hvap	85.75	kJ/mol	Joback Method
log10ws	-8.53		Crippen Method
logp	7.723		Crippen Method
mvol	352.660	ml/mol	McGowan Method
pc	933.49	kPa	Joback Method
rinpol	2916.00		NIST Webbook
rinpol	2916.00		NIST Webbook
tb	901.77	K	Joback Method
tc	1104.50	K	Joback Method
tf	504.84	K	Joback Method
vc	1.369	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1161.73	J/molxK	901.77	Joback Method
cpg	1244.18	J/molxK	1070.71	Joback Method
cpg	1230.22	J/molxK	1036.92	Joback Method
cpg	1215.03	J/molxK	1003.14	Joback Method
cpg	1198.58	J/molxK	969.35	Joback Method
cpg	1180.82	J/molxK	935.56	Joback Method
cpg	1256.96	J/molxK	1104.50	Joback Method
dvisc	0.0000280	Paxs	901.77	Joback Method

dvisc	0.0000369	Paxs	835.61	Joback Method
dvisc	0.0000509	Paxs	769.46	Joback Method
dvisc	0.0000747	Paxs	703.30	Joback Method
dvisc	0.0001187	Paxs	637.15	Joback Method
dvisc	0.0002100	Paxs	571.00	Joback Method
dvisc	0.0004314	Paxs	504.84	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U299954&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U299954&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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