

# 1H,1H,5H-Octafluoropentyl methacrylate

<b>Other names:</b>	2,2,3,3,4,4,5,5-octafluoropentyl methacrylate 2-Propenoic acid, 2-methyl-, 2,2,3,3,4,4,5,5-octafluoropentyl ester
<b>Inchi:</b>	InChI=1S/C9H8F8O2/c1-4(2)5(18)19-3-7(12,13)9(16,17)8(14,15)6(10)11/h6H,1,3H2,2H3
<b>InchiKey:</b>	ZNJXRXXJPIFFAO-UHFFFAOYSA-N
<b>Formula:</b>	C9H8F8O2
<b>SMILES:</b>	C=C(C)C(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)F
<b>Mol. weight [g/mol]:</b>	300.15
<b>CAS:</b>	355-93-1

## Physical Properties

Property code	Value	Unit	Source
gf	-1682.13	kJ/mol	Joback Method
hf	-1958.66	kJ/mol	Joback Method
hfus	18.14	kJ/mol	Joback Method
hvap	33.38	kJ/mol	Joback Method
log10ws	-3.56		Crippen Method
logp	3.277		Crippen Method
mvol	154.970	ml/mol	McGowan Method
pc	1916.94	kPa	Joback Method
tb	462.20	K	Joback Method
tc	611.30	K	Joback Method
tf	244.61	K	Joback Method
vc	0.650	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	379.54	J/molxK	462.20	Joback Method
cpg	391.37	J/molxK	487.05	Joback Method
cpg	402.50	J/molxK	511.90	Joback Method
cpg	412.98	J/molxK	536.75	Joback Method
cpg	422.83	J/molxK	561.60	Joback Method
cpg	432.08	J/molxK	586.45	Joback Method
cpg	440.76	J/molxK	611.30	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	353.00	K	3.10	NIST Webbook

## Sources

High-Pressure Phase Behavior of Binary Mixtures of Octafluoropentyl Acrylate and Octafluoropentyl Methacrylate in Supercritical Carbon Dioxide	<a href="https://www.doi.org/10.1021/je200592k">https://www.doi.org/10.1021/je200592k</a>
Joback Method	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C355931&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C355931&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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