

# Isopimara-8,15-diene

<b>Other names:</b>	8,15-Isopimaradiene
<b>Inchi:</b>	InChI=1S/C20H32/c1-6-19(4)13-10-16-15(14-19)8-9-17-18(2,3)11-7-12-20(16,17)5/h6,17
<b>InchiKey:</b>	OROJBMPJDLLRFD-MFUMQWNRSA-N
<b>Formula:</b>	C20H32
<b>SMILES:</b>	<chem>C=CC1(C)CCC2=C(CCC3C(C)(C)CCCC23C)C1</chem>
<b>Mol. weight [g/mol]:</b>	272.47

## Physical Properties

Property code	Value	Unit	Source
gf	313.63	kJ/mol	Joback Method
hf	-82.88	kJ/mol	Joback Method
hfus	12.80	kJ/mol	Joback Method
hvap	57.90	kJ/mol	Joback Method
log10ws	-6.62		Crippen Method
logp	6.286		Crippen Method
mcvol	251.480	ml/mol	McGowan Method
pc	1635.13	kPa	Joback Method
rinpol	1906.00		NIST Webbook
rinpol	1921.00		NIST Webbook
rinpol	1908.00		NIST Webbook
rinpol	1906.00		NIST Webbook
rinpol	1915.00		NIST Webbook
rinpol	1898.00		NIST Webbook
ripol	2175.00		NIST Webbook
ripol	2165.00		NIST Webbook
ripol	2175.00		NIST Webbook
ripol	2210.00		NIST Webbook
tb	700.42	K	Joback Method
tc	940.52	K	Joback Method
tf	442.88	K	Joback Method
vc	0.947	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	761.41	J/mol×K	700.42	Joback Method
cpg	787.70	J/mol×K	740.44	Joback Method
cpg	813.29	J/mol×K	780.45	Joback Method
cpg	838.65	J/mol×K	820.47	Joback Method
cpg	864.25	J/mol×K	860.49	Joback Method
cpg	890.55	J/mol×K	900.51	Joback Method
cpg	918.02	J/mol×K	940.52	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R27338&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R27338&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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