

# Hexane, 2-bromo-

<b>Other names:</b>	(dl) 2-bromohexane 2-Bromohexane
<b>Inchi:</b>	InChI=1S/C6H13Br/c1-3-4-5-6(2)7/h6H,3-5H2,1-2H3
<b>InchiKey:</b>	NEBYCXAKZCQWAW-UHFFFAOYSA-N
<b>Formula:</b>	C6H13Br
<b>SMILES:</b>	CCCCC(C)Br
<b>Mol. weight [g/mol]:</b>	165.07
<b>CAS:</b>	3377-86-4

## Physical Properties

Property code	Value	Unit	Source
gf	11.52	kJ/mol	Joback Method
hf	-146.12	kJ/mol	Joback Method
hfus	13.06	kJ/mol	Joback Method
hvap	35.00	kJ/mol	Joback Method
log10ws	-2.88		Crippen Method
logp	2.960		Crippen Method
mcvol	112.900	ml/mol	McGowan Method
pc	3392.03	kPa	Joback Method
rinpol	952.00		NIST Webbook
rinpol	934.00		NIST Webbook
rinpol	941.00		NIST Webbook
ripol	1069.00		NIST Webbook
ripol	1077.00		NIST Webbook
ripol	1091.00		NIST Webbook
tb	402.40	K	Joback Method
tc	590.50	K	Joback Method
tf	202.18	K	Joback Method
vc	0.427	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	256.62	J/molxK	590.50	Joback Method

cpg	247.93	J/molxK	559.15	Joback Method
cpg	238.81	J/molxK	527.80	Joback Method
cpg	229.23	J/molxK	496.45	Joback Method
cpg	219.19	J/molxK	465.10	Joback Method
cpg	208.66	J/molxK	433.75	Joback Method
cpg	197.64	J/molxK	402.40	Joback Method
dvisc	0.0063698	Paxs	202.18	Joback Method
dvisc	0.0003340	Paxs	402.40	Joback Method
dvisc	0.0004371	Paxs	369.03	Joback Method
dvisc	0.0006036	Paxs	335.66	Joback Method
dvisc	0.0008951	Paxs	302.29	Joback Method
dvisc	0.0014638	Paxs	268.92	Joback Method
dvisc	0.0027514	Paxs	235.55	Joback Method
hvapt	43.80	kJ/mol	359.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.50246e+01
Coeff. B	-3.78078e+03
Coeff. C	-5.70900e+01
Temperature range (K), min.	313.64
Temperature range (K), max.	446.33

## Sources

**The Yaws Handbook of Vapor Pressure:**  
**Crippen Method:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C3377864&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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