

# Diethylmalonic acid, ethyl 2-isopropoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C18H26O5/c1-6-18(7-2,16(19)21-8-3)17(20)23-15-12-10-9-11-14(15)22-13(4)5
<b>InchiKey:</b>	RMAZEWAIEYVIFNS-UHFFFAOYSA-N
<b>Formula:</b>	C18H26O5
<b>SMILES:</b>	CCOC(=O)C(CC)(CC)C(=O)Oc1ccccc1OC(C)C
<b>Mol. weight [g/mol]:</b>	322.40

## Physical Properties

Property code	Value	Unit	Source
gf	-368.98	kJ/mol	Joback Method
hf	-825.64	kJ/mol	Joback Method
hfus	31.85	kJ/mol	Joback Method
hvap	77.64	kJ/mol	Joback Method
log10ws	-4.41		Crippen Method
logp	3.749		Crippen Method
mcvol	261.470	ml/mol	McGowan Method
pc	1564.75	kPa	Joback Method
rinpol	1991.00		NIST Webbook
rinpol	1991.00		NIST Webbook
tb	814.23	K	Joback Method
tc	1022.96	K	Joback Method
tf	485.53	K	Joback Method
vc	0.985	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	791.91	J/molxK	814.23	Joback Method
cpg	807.53	J/molxK	849.02	Joback Method
cpg	821.97	J/molxK	883.81	Joback Method
cpg	835.26	J/molxK	918.59	Joback Method
cpg	847.41	J/molxK	953.38	Joback Method
cpg	858.44	J/molxK	988.17	Joback Method
cpg	868.39	J/molxK	1022.96	Joback Method
dvisc	0.0005362	Paxs	485.53	Joback Method

dvisc	0.0002792	Paxs	540.31	Joback Method
dvisc	0.0001640	Paxs	595.10	Joback Method
dvisc	0.0001053	Paxs	649.88	Joback Method
dvisc	0.0000725	Paxs	704.66	Joback Method
dvisc	0.0000526	Paxs	759.45	Joback Method
dvisc	0.0000399	Paxs	814.23	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369582&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369582&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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