

# Carbamic acid, 1-methylethyl ester

<b>Other names:</b>	Carbamic acid, isopropyl ester Isopropyl carbamate Isopropylester kyseliny karbaminove
<b>Inchi:</b>	InChI=1S/C4H9NO2/c1-3(2)7-4(5)6/h3H,1-2H3,(H2,5,6)
<b>InchiKey:</b>	OVPLZYJGTGDFNB-UHFFFAOYSA-N
<b>Formula:</b>	C4H9NO2
<b>SMILES:</b>	CC(C)OC(N)=O
<b>Mol. weight [g/mol]:</b>	103.12
<b>CAS:</b>	1746-77-6

## Physical Properties

Property code	Value	Unit	Source
gf	-187.11	kJ/mol	Joback Method
hf	-342.18	kJ/mol	Joback Method
hfus	10.58	kJ/mol	Joback Method
hvap	43.91	kJ/mol	Joback Method
log10ws	-0.88		Crippen Method
logp	0.490		Crippen Method
mcvol	84.640	ml/mol	McGowan Method
pc	4498.26	kPa	Joback Method
tb	439.30	K	Joback Method
tc	638.45	K	Joback Method
tf	275.26	K	Joback Method
vc	0.306	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	172.46	J/molxK	439.30	Joback Method
cpg	180.72	J/molxK	472.49	Joback Method
cpg	188.69	J/molxK	505.68	Joback Method
cpg	196.36	J/molxK	538.88	Joback Method
cpg	203.72	J/molxK	572.07	Joback Method
cpg	210.77	J/molxK	605.26	Joback Method

cpg

217.51

J/mol×K

638.45

Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	454.20	K	94.80	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1746776&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1746776&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/68-507-4/Carbamic-acid-1-methylethyl-ester.pdf>

Generated by Cheméo on 2024-04-17 03:49:04.774651674 +0000 UTC m=+15614993.695228989.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.