

# Pentane, 1-(1-ethoxyethoxy)-

<b>Other names:</b>	Acetaldehyde ethyl amyl acetal 1-(1-Ethoxyethoxy)pentane 4-Methyl-3,50dioxadecane 1-(1-ethoxyethoxy)pentane (acetaldehyde ethylamyl acetal)
<b>Inchi:</b>	InChI=1S/C9H20O2/c1-4-6-7-8-11-9(3)10-5-2/h9H,4-8H2,1-3H3
<b>InchiKey:</b>	QMLYOIJQQWWNKE-UHFFFAOYSA-N
<b>Formula:</b>	C9H20O2
<b>SMILES:</b>	CCCCCOC(C)OCC
<b>Mol. weight [g/mol]:</b>	160.25
<b>CAS:</b>	13442-89-2

## Physical Properties

Property code	Value	Unit	Source
gf	-187.54	kJ/mol	Joback Method
hf	-498.81	kJ/mol	Joback Method
hfus	17.92	kJ/mol	Joback Method
hvap	40.06	kJ/mol	Joback Method
log10ws	-2.37		Crippen Method
logp	2.576		Crippen Method
mcvol	149.410	ml/mol	McGowan Method
pc	2250.40	kPa	Joback Method
rinpol	966.20		NIST Webbook
rinpol	966.20		NIST Webbook
rinpol	977.00		NIST Webbook
rinpol	977.00		NIST Webbook
ripol	1104.00		NIST Webbook
ripol	1109.00		NIST Webbook
ripol	1091.00		NIST Webbook
ripol	1104.00		NIST Webbook
ripol	1091.00		NIST Webbook
ripol	1109.00		NIST Webbook
tb	449.72	K	Joback Method
tc	616.59	K	Joback Method
tf	220.65	K	Joback Method
vc	0.570	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	324.83	J/molxK	449.72	Joback Method
cpg	338.74	J/molxK	477.53	Joback Method
cpg	352.24	J/molxK	505.34	Joback Method
cpg	365.32	J/molxK	533.15	Joback Method
cpg	377.99	J/molxK	560.97	Joback Method
cpg	390.23	J/molxK	588.78	Joback Method
cpg	402.06	J/molxK	616.59	Joback Method
dvisc	0.0048122	Paxs	220.65	Joback Method
dvisc	0.0018379	Paxs	258.83	Joback Method
dvisc	0.0008990	Paxs	297.01	Joback Method
dvisc	0.0005175	Paxs	335.19	Joback Method
dvisc	0.0003336	Paxs	373.36	Joback Method
dvisc	0.0002332	Paxs	411.54	Joback Method
dvisc	0.0001733	Paxs	449.72	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13442892&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13442892&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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