

# o-Chloroacetanilide

<b>Other names:</b>	2'-Chloro acetanilide 2-Chloroacetanilide Acetamide, N-(2-chlorophenyl)- Acetanilide, 2'-chloro- N-(2-Chlorophenyl)acetamide
<b>Inchi:</b>	InChI=1S/C8H8ClNO/c1-6(11)10-8-5-3-2-4-7(8)9/h2-5H,1H3,(H,10,11)
<b>InchiKey:</b>	KNVQTRVKSOEHPU-UHFFFAOYSA-N
<b>Formula:</b>	C8H8ClNO
<b>SMILES:</b>	CC(=O)Nc1ccccc1Cl
<b>Mol. weight [g/mol]:</b>	169.61
<b>CAS:</b>	533-17-5

## Physical Properties

Property code	Value	Unit	Source
gf	67.80	kJ/mol	Joback Method
hf	-58.24	kJ/mol	Joback Method
hfus	21.02	kJ/mol	Joback Method
hvap	53.91	kJ/mol	Joback Method
ie	8.07 ± 0.03	eV	NIST Webbook
ie	8.55	eV	NIST Webbook
log10ws	-1.40		Aqueous Solubility Prediction Method
logp	2.298		Crippen Method
mcvol	123.610	ml/mol	McGowan Method
pc	3759.17	kPa	Joback Method
tb	555.57	K	Joback Method
tc	784.08	K	Joback Method
tf	359.90 ± 0.50	K	NIST Webbook
vc	0.466	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.93	J/mol×K	555.57	Joback Method

cpg	268.79	J/mol×K	593.65	Joback Method
cpg	278.91	J/mol×K	631.74	Joback Method
cpg	288.31	J/mol×K	669.82	Joback Method
cpg	297.04	J/mol×K	707.91	Joback Method
cpg	305.11	J/mol×K	745.99	Joback Method
cpg	312.56	J/mol×K	784.08	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C533175&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C533175&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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