

2,6-Pyridinediamine

Other names:	2,6-diaminopyridine DAP NSC 1921 Pyridine, 2,6-diamino- pyridine-2,6-diamine pyridine-2,6-diylldiamine
Inchi:	InChI=1S/C5H7N3/c6-4-2-1-3-5(7)8-4/h1-3H,(H4,6,7,8)
InchiKey:	VHNQIURBCCNWDN-UHFFFAOYSA-N
Formula:	C5H7N3
SMILES:	<chem>Nc1cccc(N)n1</chem>
Mol. weight [g/mol]:	109.13
CAS:	141-86-6

Physical Properties

Property code	Value	Unit	Source
chs	-2961.40 ± 0.46	kJ/mol	NIST Webbook
log10ws	-0.50		Crippen Method
logp	0.246		Crippen Method
mvol	87.490	ml/mol	McGowan Method
rmpol	1377.00		NIST Webbook
tb	558.20	K	NIST Webbook
tf	394.80	K	Liquid-Liquid Equilibria for Systems of 1-Butanol + Water + 2,6-Diaminopyridine and 1-Butanol + Water + 2-Aminopyridine

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	422.20	K	0.70	NIST Webbook

Sources

Liquid-Liquid Equilibria for Systems of 1-Butanol + Water + 2,6-Diaminopyridine and 1-Butanol + Water + 2-Aminopyridine: <https://www.doi.org/10.1021/je8007843>
McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>
NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C141866&Units=SI>
Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>
Crippen Method: https://www.cheméo.com/doc/models/crippen_log10ws
Solubility of 2,6-Diaminopyridine in Toluene, o-Xylene, Ethylbenzene, Methanol, Ethanol, 2-Propanol, and Sodium Hydroxide Solutions: <https://www.doi.org/10.1021/je700221w>

Legend

chs: Standard solid enthalpy of combustion
log10ws: Log10 of Water solubility in mol/l
logp: Octanol/Water partition coefficient
mcvol: McGowan's characteristic volume
rinpol: Non-polar retention indices
tb: Normal Boiling Point Temperature
tbrp: Boiling point at reduced pressure
tf: Normal melting (fusion) point

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