

# Niflumic acid

<b>Other names:</b>	2-(3-(Trifluoromethyl)-phenyl)aminonicotinic acid 2-[3-(Trifluoromethyl)anilino]-3-pyridinecarboxylic acid 2-[3-(Trifluoromethyl)anilino]nicotinic acid 2-[3-(Trifluoromethyl)anilino]nicotinic acid (niflumic acid) 2-[[3-(trifluoromethyl)phenyl]amino]-3-pyridinecarboxylic acid 3- pyridinecarboxylic acid, 2-[[3-(trifluoromethyl)phenyl]amino]- 3-Pyridinecarboxylic acid, 2-[[3-(trifluoromethyl)phenyl]amino]- Acide niflumique Acido niflumico Actol Actol, analgesic Flogovital Forenol Landruma Nicotinic acid, 2-(«alpha», «alpha», «alpha»-trifluoro-m-toluidino)- Nicotinic acid, 2-(À«alphaÂ», Â«alphaÂ», Â«alphaÂ»-trifluoro-m-toluidino)- Niflamol Nifluminic acid Nifluril UP 83
<b>Inchi:</b>	InChI=1S/C13H9F3N2O2/c14-13(15,16)8-3-1-4-9(7-8)18-11-10(12(19)20)5-2-6-17-11/h1
<b>InchiKey:</b>	JZFPYUNJRRFVQU-UHFFFAOYSA-N
<b>Formula:</b>	C13H9F3N2O2
<b>SMILES:</b>	O=C(O)c1cccn1Nc1cccc(C(F)(F)F)c1
<b>Mol. weight [g/mol]:</b>	282.22
<b>CAS:</b>	4394-00-7

## Physical Properties

Property code	Value	Unit	Source
hsub	130.20 ± 0.80	kJ/mol	NIST Webbook
hvap	107.50	kJ/mol	NIST Webbook
log10ws	-4.00		Aqueous Solubility Prediction Method
logp	3.542		Crippen Method
mcvol	179.220	ml/mol	McGowan Method
rinpol	2093.00		NIST Webbook

tf	477.15	K	Determination, correlation and prediction of the solubilities of niflumic acid, clofenamic acid and tolfenamic acid in supercritical CO <sub>2</sub>
tf	476.00 ± 0.10	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hfust	36.50	kJ/mol	478.00	NIST Webbook
hfust	35.70	kJ/mol	476.40	NIST Webbook
hfust	32.73	kJ/mol	477.20	NIST Webbook
hfust	38.00	kJ/mol	476.00	NIST Webbook
hsubt	127.80 ± 0.80	kJ/mol	375.50	NIST Webbook

## Sources

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C4394007&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

**Determination, correlation and prediction of the solubilities of niflumic Solubilities of Organic Semiconductors and Nonsteroidal Anti-inflammatory Drugs in Pure and Mixed Organic Solvents: Measurement and Modeling With Hansen Solubility Parameter:**

<https://www.doi.org/10.1016/j.fluid.2015.02.007>

<https://www.doi.org/10.1021/acs.jced.8b00536>

**Gibbs Free Energy and Mixed Organic Solvents: Measurement and Modeling With Hansen Solubility Parameter:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

<http://link.springer.com/article/10.1007/BF02311772>

## Legend

- hfust:** Enthalpy of fusion at a given temperature
- hsub:** Enthalpy of sublimation at standard conditions
- hsubt:** Enthalpy of sublimation at a given temperature
- hvap:** Enthalpy of vaporization at standard conditions
- log10ws:** Log10 of Water solubility in mol/l
- logp:** Octanol/Water partition coefficient
- mcvol:** McGowan's characteristic volume
- rinpol:** Non-polar retention indices
- tf:** Normal melting (fusion) point

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