

3,5-Bis(trifluoromethyl)benzenesulfonyl chloride

Other names:	3,5-Di(trifluoromethyl)benzene sulfonyl chloride Benzenesulfonyl chloride, 3,5-bis(trifluoromethyl)- 3,5-bis(trifluoromethyl)benzenesulphonyl chloride
Inchi:	InChI=1S/C8H3ClF6O2S/c9-18(16,17)6-2-4(7(10,11)12)1-5(3-6)8(13,14)15/h1-3H
InchiKey:	BTRCVKADYDVSLI-UHFFFAOYSA-N
Formula:	C8H3ClF6O2S
SMILES:	O=S(=O)(Cl)c1cc(C(F)(F)F)cc(C(F)(F)F)c1
Mol. weight [g/mol]:	312.62
CAS:	39234-86-1

Physical Properties

Property code	Value	Unit	Source
gf	-1534.02	kJ/mol	Joback Method
hf	-1658.11	kJ/mol	Joback Method
hfus	28.97	kJ/mol	Joback Method
hvap	52.53	kJ/mol	Joback Method
log10ws	-4.10		Crippen Method
logp	3.652		Crippen Method
mcvol	150.770	ml/mol	McGowan Method
pc	3089.85	kPa	Joback Method
tb	493.45	K	Joback Method
tc	670.31	K	Joback Method
tf	308.24	K	Joback Method
vc	0.636	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	332.54	J/molxK	493.45	Joback Method
cpg	342.96	J/molxK	522.93	Joback Method
cpg	352.65	J/molxK	552.40	Joback Method
cpg	361.65	J/molxK	581.88	Joback Method
cpg	369.98	J/molxK	611.36	Joback Method
cpg	377.66	J/molxK	640.83	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C39234861&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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