

# Benzene, 1-chloro-4-methyl-2-nitro-

<b>Other names:</b>	Toluene, 4-chloro-3-nitro- 3-Nitro-4-chlorotoluene 4-Chloro-3-nitrotoluene 4,3-Chloronitrotoluene 1-Chloro-4-methyl-2-nitrobenzene 2-Chloro-5-methylnitrobenzene
<b>Inchi:</b>	InChI=1S/C7H6ClNO2/c1-5-2-3-6(8)7(4-5)9(10)11/h2-4H,1H3
<b>InchiKey:</b>	NWESJZZPAJGHRZ-UHFFFAOYSA-N
<b>Formula:</b>	C7H6ClNO2
<b>SMILES:</b>	<chem>Cc1ccc(Cl)c([N+](=O)[O-])c1</chem>
<b>Mol. weight [g/mol]:</b>	171.58
<b>CAS:</b>	89-60-1

## Physical Properties

Property code	Value	Unit	Source
gf	124.83	kJ/mol	Joback Method
hf	-0.72	kJ/mol	Joback Method
hfus	22.71	kJ/mol	Joback Method
hvap	55.75	kJ/mol	Joback Method
log10ws	-3.28		Crippen Method
logp	2.557		Crippen Method
mcvol	115.390	ml/mol	McGowan Method
pc	3838.78	kPa	Joback Method
rinpol	1312.00		NIST Webbook
rinpol	1312.00		NIST Webbook
ripol	2039.00		NIST Webbook
ripol	2039.00		NIST Webbook
tb	585.47	K	Joback Method
tc	840.46	K	Joback Method
tf	280.33 ± 0.30	K	NIST Webbook
tf	280.40 ± 0.20	K	NIST Webbook
tf	279.90 ± 0.25	K	NIST Webbook
tf	280.11 ± 0.35	K	NIST Webbook
vc	0.451	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	241.86	J/molxK	585.47	Joback Method
cpg	251.78	J/molxK	627.97	Joback Method
cpg	260.95	J/molxK	670.47	Joback Method
cpg	269.40	J/molxK	712.97	Joback Method
cpg	277.17	J/molxK	755.46	Joback Method
cpg	284.29	J/molxK	797.96	Joback Method
cpg	290.80	J/molxK	840.46	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	533.20	K	99.30	NIST Webbook
tbrp	391.20	K	1.50	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C89601&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C89601&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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