

# 1-Pentene, 2,4-dimethyl-

Other names:

(CH3)2CHCH2C(CH3)=CH2  
2,4-Dimethyl-1-pentene  
2,4-dimethylpent-1-ene

Inchi:

InChI=1S/C7H14/c1-6(2)5-7(3)4/h7H,1,5H2,2-4H3

InchiKey:

LXQPBCHJNIOMQU-UHFFFAOYSA-N

Formula:

C7H14

SMILES:

C=C(C)CC(C)C

Mol. weight [g/mol]:

98.19

CAS:

2213-32-3

## Physical Properties

| Property code | Value           | Unit   | Source         |
|---------------|-----------------|--------|----------------|
| chl           | -4638.30 ± 1.70 | kJ/mol | NIST Webbook   |
| gf            | 84.91           | kJ/mol | Joback Method  |
| hf            | -77.45          | kJ/mol | Joback Method  |
| hfus          | 7.77            | kJ/mol | Joback Method  |
| hvap          | 33.10           | kJ/mol | NIST Webbook   |
| hvap          | 33.10           | kJ/mol | NIST Webbook   |
| ie            | 9.03 ± 0.01     | eV     | NIST Webbook   |
| log10ws       | -2.36           |        | Crippen Method |
| logp          | 2.609           |        | Crippen Method |
| mcvol         | 105.190         | ml/mol | McGowan Method |
| pc            | 2969.80         | kPa    | Joback Method  |
| rinpol        | 636.00          |        | NIST Webbook   |
| rinpol        | 638.00          |        | NIST Webbook   |
| rinpol        | 639.80          |        | NIST Webbook   |
| rinpol        | 638.00          |        | NIST Webbook   |
| rinpol        | 639.00          |        | NIST Webbook   |
| rinpol        | 639.00          |        | NIST Webbook   |
| rinpol        | 636.00          |        | NIST Webbook   |
| rinpol        | 638.00          |        | NIST Webbook   |
| rinpol        | 640.00          |        | NIST Webbook   |
| rinpol        | 641.00          |        | NIST Webbook   |
| rinpol        | 640.00          |        | NIST Webbook   |
| rinpol        | 637.50          |        | NIST Webbook   |
| rinpol        | 634.00          |        | NIST Webbook   |
| rinpol        | 646.60          |        | NIST Webbook   |

|        |               |         |               |
|--------|---------------|---------|---------------|
| rinpol | 656.00        |         | NIST Webbook  |
| rinpol | 648.10        |         | NIST Webbook  |
| rinpol | 646.00        |         | NIST Webbook  |
| rinpol | 636.90        |         | NIST Webbook  |
| rinpol | 636.00        |         | NIST Webbook  |
| rinpol | 638.00        |         | NIST Webbook  |
| rinpol | 638.00        |         | NIST Webbook  |
| rinpol | 642.00        |         | NIST Webbook  |
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| rinpol | 646.00        |         | NIST Webbook  |
| rinpol | 639.20        |         | NIST Webbook  |
| rinpol | 637.90        |         | NIST Webbook  |
| rinpol | 645.30        |         | NIST Webbook  |
| rinpol | 647.00        |         | NIST Webbook  |
| rinpol | 646.00        |         | NIST Webbook  |
| rinpol | 645.00        |         | NIST Webbook  |
| rinpol | 636.20        |         | NIST Webbook  |
| rinpol | 646.50        |         | NIST Webbook  |
| rinpol | 657.00        |         | NIST Webbook  |
| rinpol | 645.90        |         | NIST Webbook  |
| ripol  | 734.00        |         | NIST Webbook  |
| tb     | 355.68        | K       | Joback Method |
| tc     | 530.11        | K       | Joback Method |
| tf     | 149.17 ± 0.50 | K       | NIST Webbook  |
| tf     | 149.06 ± 0.06 | K       | NIST Webbook  |
| tf     | 149.06 ± 0.04 | K       | NIST Webbook  |
| tf     | 149.09 ± 0.02 | K       | NIST Webbook  |
| tf     | 148.95 ± 0.50 | K       | NIST Webbook  |
| vc     | 0.404         | m3/kmol | Joback Method |

# Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 180.34 | J/mol×K | 355.68          | Joback Method |
| cpg           | 192.16 | J/mol×K | 384.75          | Joback Method |
| cpg           | 203.51 | J/mol×K | 413.82          | Joback Method |
| cpg           | 214.40 | J/mol×K | 442.89          | Joback Method |
| cpg           | 224.84 | J/mol×K | 471.96          | Joback Method |
| cpg           | 234.84 | J/mol×K | 501.04          | Joback Method |
| cpg           | 244.42 | J/mol×K | 530.11          | Joback Method |
| hvapt         | 32.30  | kJ/mol  | 336.00          | NIST Webbook  |
| hvapt         | 33.20  | kJ/mol  | 322.00          | NIST Webbook  |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.39574e+01                   |
| Coeff. B                    | -2.86746e+03                  |
| Coeff. C                    | -4.77230e+01                  |
| Temperature range (K), min. | 257.49                        |
| Temperature range (K), max. | 379.38                        |

| Information                 | Value                                      |
|-----------------------------|--------------------------------------------|
| Property code               | pvap                                       |
| Equation                    | $\ln(P_{vp}) = A + B/T + C*\ln(T) + D*T^2$ |
| Coeff. A                    | 8.26910e+01                                |
| Coeff. B                    | -6.60401e+03                               |
| Coeff. C                    | -1.03271e+01                               |
| Coeff. D                    | 9.35835e-06                                |
| Temperature range (K), min. | 289.15                                     |
| Temperature range (K), max. | 355.15                                     |

# Sources

|                                             |                                                                                                                                                                                         |
|---------------------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Joback Method:</b>                       | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                                                                   |
| <b>KDB:</b>                                 | <a href="https://www.cheric.org/files/research/kdb/mol/mol236.mol">https://www.cheric.org/files/research/kdb/mol/mol236.mol</a>                                                         |
| <b>McGowan Method:</b>                      | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                                                                   |
| <b>NIST Webbook:</b>                        | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2213323&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2213323&amp;Units=SI</a>                                             |
| <b>The Yaws Handbook of Vapor Pressure:</b> | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |
| <b>KDB Vapor Pressure Data:</b>             | <a href="https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=236">https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=236</a>                                       |
| <b>Crippen Method:</b>                      | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                                                               |
| <b>Crippen Method:</b>                      | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                                                                       |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>chl:</b>     | Standard liquid enthalpy of combustion          |
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>hvapt:</b>   | Enthalpy of vaporization at a given temperature |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>pvap:</b>    | Vapor pressure                                  |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>ripol:</b>   | Polar retention indices                         |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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