

# Benzenamine, 4-methoxy-N-(phenylmethylene)-

<b>Other names:</b>	p-Anisidine, N-benzylidene- Benzylidene-p-methoxyaniline N-Benzylidene-p-methoxyaniline N-Benzylidene-4-methoxyaniline N-Benzyliden-p-anisidin N-Benzylidene-p-anisidine 1-Methoxybenzene, 4-benzylidenamino 4-Methoxy-N-[phenylmethylidene]aniline Benzylidene-(4-methoxyphenyl)-amine
<b>Inchi:</b>	InChI=1S/C14H13NO/c1-16-14-9-7-13(8-10-14)15-11-12-5-3-2-4-6-12/h2-11H,1H3
<b>InchiKey:</b>	LKHMZCUKGPUKEA-UHFFFAOYSA-N
<b>Formula:</b>	C14H13NO
<b>SMILES:</b>	<chem>COc1ccc(N=Cc2ccccc2)cc1</chem>
<b>Mol. weight [g/mol]:</b>	211.26
<b>CAS:</b>	783-08-4

## Physical Properties

Property code	Value	Unit	Source
hf	79.30	kJ/mol	Joback Method
hvap	57.70	kJ/mol	Joback Method
log10ws	-3.46		Crippen Method
logp	3.446		Crippen Method
mcvol	172.150	ml/mol	McGowan Method
pc	2410.00	kPa	Joback Method
rinpol	2035.00		NIST Webbook
rinpol	2035.00		NIST Webbook
tb	677.16	K	Joback Method
tc	929.02	K	Joback Method

## Sources

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C783084&Units=SI>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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