

# Behenic alcohol

<b>Other names:</b>	1-Docosanol Abreva Behenyl alcohol Cachalot BE-22 Dehydag wax 22 (lanette) Docosan-1-ol Docosanol Docosanol-(1) Docosyl alcohol Emery 3304 IK 2 Lanette 22 Loxiol VPG 1451 N-Eicosanol NAA 422 NSC 8407 Nacol 22-97 Stenol 1822 Stenol 1822A Tadenan n-Docosanol
<b>Inchi:</b>	InChI=1S/C22H46O/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23/h23
<b>InchiKey:</b>	NOPFSRXAKWQILS-UHFFFAOYSA-N
<b>Formula:</b>	C22H46O
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCCO
<b>Mol. weight [g/mol]:</b>	326.60
<b>CAS:</b>	661-19-8

## Physical Properties

Property code	Value	Unit	Source
chl	-14450.00 ± 10.00	kJ/mol	NIST Webbook
gf	-2.46	kJ/mol	Joback Method
hf	-649.64	kJ/mol	Joback Method
hfus	56.82	kJ/mol	Joback Method
hsub	207.00 ± 10.00	kJ/mol	NIST Webbook
hsub	239.00 ± 10.00	kJ/mol	NIST Webbook

hvap	135.90 ± 0.80	kJ/mol	NIST Webbook
log10ws	-8.29		Crippen Method
logp	7.801		Crippen Method
mcvol	326.710	ml/mol	McGowan Method
pc	953.78	kPa	Joback Method
rinpol	2468.00		NIST Webbook
rinpol	2473.00		NIST Webbook
rinpol	2498.00		NIST Webbook
rinpol	2456.00		NIST Webbook
rinpol	397.16		NIST Webbook
rinpol	401.29		NIST Webbook
rinpol	2476.00		NIST Webbook
rinpol	400.57		NIST Webbook
rinpol	367.90		NIST Webbook
rinpol	2470.40		NIST Webbook
rinpol	2475.00		NIST Webbook
rinpol	2470.40		NIST Webbook
rinpol	2456.00		NIST Webbook
rinpol	2501.00		NIST Webbook
rinpol	2501.40		NIST Webbook
rinpol	2470.00		NIST Webbook
rinpol	2456.00		NIST Webbook
rinpol	2493.00		NIST Webbook
rinpol	2470.00		NIST Webbook
rinpol	2494.00		NIST Webbook
rinpol	2443.00		NIST Webbook
rinpol	2497.00		NIST Webbook
rinpol	2467.00		NIST Webbook
rinpol	2456.00		NIST Webbook
rinpol	2501.40		NIST Webbook
ripol	2979.00		NIST Webbook
ripol	2979.00		NIST Webbook
tb	794.94	K	Joback Method
tc	973.24	K	Joback Method
tf	344.50	K	Evaluation of the Vaporization, Fusion, and Sublimation Enthalpies of the 1-Alkanols: The Vaporization Enthalpy of 1-, 6-, 7-, and 9-Heptadecanol, 1-Octadecanol, 1-Eicosanol, 1-Docosanol, 1-Hexacosanol, and Cholesterol at T ) 298.15 K by Correlation Gas Chromatography
tf	343.20 ± 3.00	K	NIST Webbook

tf	342.40 ± 0.60	K	NIST Webbook
tf	343.00 ± 0.25	K	NIST Webbook
vc	1.286	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1050.02	J/mol×K	794.94	Joback Method
cpg	1159.25	J/mol×K	973.24	Joback Method
cpg	1090.11	J/mol×K	854.37	Joback Method
cpg	1108.71	J/mol×K	884.09	Joback Method
cpg	1126.41	J/mol×K	913.81	Joback Method
cpg	1143.24	J/mol×K	943.52	Joback Method
cpg	1070.56	J/mol×K	824.66	Joback Method
dvisc	0.0000216	Paxs	728.87	Joback Method
dvisc	0.0000379	Paxs	662.80	Joback Method
dvisc	0.0000752	Paxs	596.73	Joback Method
dvisc	0.0001770	Paxs	530.66	Joback Method
dvisc	0.0005314	Paxs	464.59	Joback Method
dvisc	0.0000136	Paxs	794.94	Joback Method
dvisc	0.0022967	Paxs	398.52	Joback Method
hfust	86.06	kJ/mol	343.90	NIST Webbook
hfust	85.07	kJ/mol	240.00	NIST Webbook
hfust	46.57	kJ/mol	345.20	NIST Webbook
hfust	17.24	kJ/mol	333.90	NIST Webbook
hfust	46.57	kJ/mol	345.20	NIST Webbook
hsubt	207.00 ± 10.00	kJ/mol	338.00	NIST Webbook
hvapt	115.30	kJ/mol	401.50	NIST Webbook
sfust	50.72	J/mol×K	333.90	NIST Webbook
sfust	134.90	J/mol×K	345.20	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	453.20	K	0.03	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51940e+01
Coeff. B	-5.76455e+03
Coeff. C	-1.22684e+02
Temperature range (K), min.	509.40
Temperature range (K), max.	705.99

## Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Phase equilibria of high molecular mass 1-alcohols in supercritical propane	<a href="https://www.doi.org/10.1016/j.fluid.2007.04.027">https://www.doi.org/10.1016/j.fluid.2007.04.027</a>
Evaluation of the Vaporization, Fusion, and Sublimation Enthalpies of the 40 Alkanols	<a href="https://www.doi.org/10.1021/je0503857">https://www.doi.org/10.1021/je0503857</a>
Joback Method: The Vaporization Enthalpy of 1-, 6-, 7- and 9-Heptadecanol, 1-Octadecanol, 1-Eicosanol, 1-Decosanol, 1-Hexacosanol, and Cholesterol at T ) 298.15 K by the Joback Method	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C661198&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C661198&amp;Units=SI</a>
Correlation Gas Chromatography: The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hsub:	Enthalpy of sublimation at standard conditions
hsubt:	Enthalpy of sublimation at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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