

Hexane, 2,4,4-trimethyl-

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|-----------------------------|---|
| Other names: | 2,4,4-Trimethylhexane |
| Inchi: | InChI=1S/C9H20/c1-6-9(4,5)7-8(2)3/h8H,6-7H2,1-5H3 |
| InchiKey: | SVEMKBCPZYWEPH-UHFFFAOYSA-N |
| Formula: | C9H20 |
| SMILES: | CCC(C)(C)CC(C)C |
| Mol. weight [g/mol]: | 128.26 |
| CAS: | 16747-30-1 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|-----------------|--------|----------------|
| chl | -6119.73 ± 0.75 | kJ/mol | NIST Webbook |
| gf | 25.30 | kJ/mol | Joback Method |
| hf | -243.12 | kJ/mol | Joback Method |
| hfl | -280.20 ± 0.92 | kJ/mol | NIST Webbook |
| hfus | 8.13 | kJ/mol | Joback Method |
| hvap | 41.10 | kJ/mol | NIST Webbook |
| log10ws | -3.11 | | Crippen Method |
| logp | 3.469 | | Crippen Method |
| mcvol | 137.670 | ml/mol | McGowan Method |
| pc | 2367.97 | kPa | Joback Method |
| rinpol | 808.00 | | NIST Webbook |
| rinpol | 812.00 | | NIST Webbook |
| rinpol | 810.00 | | NIST Webbook |
| rinpol | 810.00 | | NIST Webbook |
| rinpol | 806.90 | | NIST Webbook |
| rinpol | 809.80 | | NIST Webbook |
| rinpol | 808.70 | | NIST Webbook |
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| rinpol | 809.70 | | NIST Webbook |
| rinpol | 808.00 | | NIST Webbook |
| rinpol | 811.00 | | NIST Webbook |
| rinpol | 808.30 | | NIST Webbook |
| rinpol | 810.00 | | NIST Webbook |
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| rinpol | 811.00 | | NIST Webbook |

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| rinpol | 810.00 | | NIST Webbook |
| rinpol | 809.00 | | NIST Webbook |
| rinpol | 808.00 | | NIST Webbook |
| rinpol | 810.10 | | NIST Webbook |
| rinpol | 812.20 | | NIST Webbook |
| rinpol | 807.42 | | NIST Webbook |
| rinpol | 806.00 | | NIST Webbook |
| rinpol | 812.00 | | NIST Webbook |
| rinpol | 808.00 | | NIST Webbook |
| rinpol | 807.00 | | NIST Webbook |
| rinpol | 816.00 | | NIST Webbook |
| rinpol | 811.00 | | NIST Webbook |
| rinpol | 806.00 | | NIST Webbook |
| rinpol | 810.00 | | NIST Webbook |
| rinpol | 805.00 | | NIST Webbook |
| tb | 401.65 | K | Joback Method |
| tc | 579.13 | K | Joback Method |
| tf | 159.77 ± 0.05 | K | NIST Webbook |
| tf | 159.77 ± 0.04 | K | NIST Webbook |
| tf | 155.45 ± 0.50 | K | NIST Webbook |
| vc | 0.522 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 273.89 | J/molxK | 401.65 | Joback Method |
| cpg | 289.98 | J/molxK | 431.23 | Joback Method |
| cpg | 305.33 | J/molxK | 460.81 | Joback Method |
| cpg | 319.96 | J/molxK | 490.39 | Joback Method |
| cpg | 333.90 | J/molxK | 519.97 | Joback Method |
| cpg | 347.18 | J/molxK | 549.55 | Joback Method |
| cpg | 359.82 | J/molxK | 579.13 | Joback Method |
| dvisc | 0.0222427 | Paxs | 178.61 | Joback Method |
| dvisc | 0.0055664 | Paxs | 215.78 | Joback Method |
| dvisc | 0.0020930 | Paxs | 252.96 | Joback Method |
| dvisc | 0.0010112 | Paxs | 290.13 | Joback Method |
| dvisc | 0.0005763 | Paxs | 327.30 | Joback Method |
| dvisc | 0.0003684 | Paxs | 364.48 | Joback Method |
| dvisc | 0.0002558 | Paxs | 401.65 | Joback Method |
| hvapt | 38.50 | kJ/mol | 364.50 | NIST Webbook |

Correlations

| Information | Value |
|-----------------------------|-------------------------------|
| Property code | pvap |
| Equation | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A | 1.41409e+01 |
| Coeff. B | -3.35035e+03 |
| Coeff. C | -5.19800e+01 |
| Temperature range (K), min. | 293.82 |
| Temperature range (K), max. | 431.43 |

| Information | Value |
|-----------------------------|--|
| Property code | pvap |
| Equation | $\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$ |
| Coeff. A | 8.66959e+01 |
| Coeff. B | -7.69356e+03 |
| Coeff. C | -1.06981e+01 |
| Coeff. D | 7.20088e-06 |
| Temperature range (K), min. | 293.15 |
| Temperature range (K), max. | 581.50 |

Sources

| | |
|---|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C16747301&Units=SI |
| The Yaws Handbook of Vapor Pressure: | https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure |
| KDB Vapor Pressure Data: | https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=88 |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| KDB: | https://www.thermo.com/files/research/kdb/mol/mol88.mol |

Legend

| | |
|-----------------|---|
| chl: | Standard liquid enthalpy of combustion |
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfl: | Liquid phase enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| hvapt: | Enthalpy of vaporization at a given temperature |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| pvap: | Vapor pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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