

# Flamenol

<b>Other names:</b>	5-Methoxyresorcinol 3,5-Dihydroxyanisole 1,3-Benzenediol, 5-methoxy-
<b>Inchi:</b>	InChI=1S/C7H8O3/c1-10-7-3-5(8)2-6(9)4-7/h2-4,8-9H,1H3
<b>InchiKey:</b>	HDVRLUFGYQYLFJ-UHFFFAOYSA-N
<b>Formula:</b>	C7H8O3
<b>SMILES:</b>	COc1cc(O)cc(O)c1
<b>Mol. weight [g/mol]:</b>	140.14
<b>CAS:</b>	2174-64-3

## Physical Properties

Property code	Value	Unit	Source
gf	-293.77	kJ/mol	Joback Method
hf	-438.12	kJ/mol	Joback Method
hfus	20.68	kJ/mol	Joback Method
hvap	61.89	kJ/mol	Joback Method
log10ws	-0.69		Crippen Method
logp	1.106		Crippen Method
mcvol	103.340	ml/mol	McGowan Method
pc	6122.63	kPa	Joback Method
tb	569.90	K	Joback Method
tc	809.70	K	Joback Method
tf	440.74	K	Joback Method
vc	0.270	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	247.31	J/mol×K	569.90	Joback Method
cpg	256.37	J/mol×K	609.87	Joback Method
cpg	264.71	J/mol×K	649.83	Joback Method
cpg	272.46	J/mol×K	689.80	Joback Method
cpg	279.72	J/mol×K	729.77	Joback Method
cpg	286.62	J/mol×K	769.73	Joback Method

cpg	293.27	J/molxK	809.70	Joback Method
dvisc	0.0002233	Paxs	440.74	Joback Method
dvisc	0.0001166	Paxs	462.27	Joback Method
dvisc	0.0000645	Paxs	483.79	Joback Method
dvisc	0.0000375	Paxs	505.32	Joback Method
dvisc	0.0000228	Paxs	526.85	Joback Method
dvisc	0.0000144	Paxs	548.37	Joback Method
dvisc	0.0000094	Paxs	569.90	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	461.70	K	1.60	NIST Webbook
tbrp	486.20	K	2.10	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2174643&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2174643&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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