

# 1,1'-Bicyclohexyl

<b>Other names:</b>	1,1'-Biphenyl, dodecahydro- 1,1-Bicyclohexyl BICYCLOHEXANE Bicyclohexyl Cyclohexane, cyclohexyl- Cyclohexylcyclohexane DICYCLOHEXYL DODECAHYDRO-1,1-BIPHENYL Dicyclohexane Dodecahydrobiphenyl NSC 59855
<b>Inchi:</b>	InChI=1S/C12H22/c1-3-7-11(8-4-1)12-9-5-2-6-10-12/h11-12H,1-10H2
<b>InchiKey:</b>	WVIIMZNLDSIRH-UHFFFAOYSA-N
<b>Formula:</b>	C12H22
<b>SMILES:</b>	C1CCC(C2CCCCC2)CC1
<b>Mol. weight [g/mol]:</b>	166.30
<b>CAS:</b>	92-51-3

## Physical Properties

Property code	Value	Unit	Source
chl	-7537.50 ± 6.30	kJ/mol	NIST Webbook
chl	-7596.50	kJ/mol	NIST Webbook
chl	-7439.65	kJ/mol	NIST Webbook
chl	-7589.90 ± 2.60	kJ/mol	NIST Webbook
chl	-7593.00 ± 1.00	kJ/mol	NIST Webbook
gf	99.06	kJ/mol	Joback Method
hf	-218.40 ± 3.10	kJ/mol	NIST Webbook
hf	-215.30	kJ/mol	NIST Webbook
hfl	-276.40 ± 3.10	kJ/mol	NIST Webbook
hfl	-273.30 ± 1.20	kJ/mol	NIST Webbook
hfus	10.51	kJ/mol	Joback Method
hvap	58.00 ± 0.20	kJ/mol	NIST Webbook
hvap	58.00	kJ/mol	NIST Webbook
hvap	58.00 ± 0.30	kJ/mol	NIST Webbook
hvap	57.98	kJ/mol	NIST Webbook
ie	9.41	eV	NIST Webbook
log10ws	-4.15		Crippen Method

logp	4.147		Crippen Method
mcvol	158.220	ml/mol	McGowan Method
pc	2603.08	kPa	Joback Method
rinpol	1313.00		NIST Webbook
rinpol	1284.00		NIST Webbook
rinpol	1327.00		NIST Webbook
rinpol	1284.00		NIST Webbook
rinpol	1295.00		NIST Webbook
rinpol	1301.20		NIST Webbook
rinpol	1293.10		NIST Webbook
rinpol	1301.50		NIST Webbook
rinpol	1307.80		NIST Webbook
rinpol	1304.39		NIST Webbook
rinpol	1293.10		NIST Webbook
rinpol	1301.50		NIST Webbook
rinpol	1307.80		NIST Webbook
rinpol	1301.50		NIST Webbook
rinpol	1301.20		NIST Webbook
rinpol	1312.00		NIST Webbook
rinpol	1298.00		NIST Webbook
rinpol	1313.00		NIST Webbook
rinpol	1309.00		NIST Webbook
rinpol	1324.00		NIST Webbook
rinpol	1327.00		NIST Webbook
rinpol	1301.50		NIST Webbook
ripol	1420.00		NIST Webbook
ripol	1431.00		NIST Webbook
tb	513.06	K	Joback Method
tc	745.11	K	Joback Method
tf	239.76	K	Joback Method
tt	276.90 ± 0.00	K	NIST Webbook
vc	0.574	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	385.05	J/mol×K	513.06	Joback Method
cpg	518.21	J/mol×K	745.11	Joback Method
cpg	499.72	J/mol×K	706.43	Joback Method
cpg	479.81	J/mol×K	667.76	Joback Method
cpg	435.56	J/mol×K	590.41	Joback Method

cpg	458.45	J/molxK	629.08	Joback Method
cpg	411.12	J/molxK	551.73	Joback Method
cpl	300.40	J/molxK	311.00	NIST Webbook
cpl	300.00	J/molxK	313.00	NIST Webbook
cpl	283.00	J/molxK	298.15	NIST Webbook
dvisc	0.0003514	Paxs	467.51	Joback Method
dvisc	0.0002550	Paxs	513.06	Joback Method
dvisc	0.0035294	Paxs	285.31	Joback Method
dvisc	0.0008424	Paxs	376.41	Joback Method
dvisc	0.0108669	Paxs	239.76	Joback Method
dvisc	0.0005190	Paxs	421.96	Joback Method
dvisc	0.0015623	Paxs	330.86	Joback Method
hfust	0.74	kJ/mol	267.50	NIST Webbook
hfust	6.86	kJ/mol	223.00	NIST Webbook
hfust	6.78	kJ/mol	277.20	NIST Webbook
hfust	1.54	kJ/mol	256.10	NIST Webbook
hfust	7.08	kJ/mol	273.50	NIST Webbook
hfust	6.78	kJ/mol	277.20	NIST Webbook
hvapt	42.50	kJ/mol	525.00	NIST Webbook
hvapt	50.10	kJ/mol	435.00	NIST Webbook
hvapt	53.80	kJ/mol	421.00	NIST Webbook
rfi	1.47870		293.15	Density, Viscosity, Refractive Index, and Freezing Point for Binary Mixtures of 1,1'-Bicyclohexyl with Alkylcyclohexane
rhol	871.74	kg/m3	313.15	Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K, Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and 1,2,3,4-Tetrahydronaphthalene or Trans-decahydronaphthalene at 0.1 MPa

rho1	864.58	kg/m3	323.15	Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K, Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and 1,2,3,4-Tetrahydronaphthalene or Trans-decahydronaphthalene at 0.1 MPa
rho1	857.43	kg/m3	333.15	Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K, Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and 1,2,3,4-Tetrahydronaphthalene or Trans-decahydronaphthalene at 0.1 MPa
rho1	878.93	kg/m3	303.15	Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K, Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and 1,2,3,4-Tetrahydronaphthalene or Trans-decahydronaphthalene at 0.1 MPa

rho1	843.00	kg/m3	353.15	Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K, Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and 1,2,3,4-Tetrahydronaphthalene or Trans-decahydronaphthalene at 0.1 MPa
rho1	835.90	kg/m3	363.15	Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K, Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and 1,2,3,4-Tetrahydronaphthalene or Trans-decahydronaphthalene at 0.1 MPa
rho1	886.13	kg/m3	293.15	Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K, Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and 1,2,3,4-Tetrahydronaphthalene or Trans-decahydronaphthalene at 0.1 MPa
rho1	886.05	kg/m3	293.15	Densities and Viscosities of Ternary System n-Dodecane (1) + Bicyclohexyl (2) + n-Butanol (3) and Corresponding Binaries at T = (293.15 to 333.15) K

rho1	882.44	kg/m3	298.15	Densities and Viscosities of Ternary System n-Dodecane (1) + Bicyclohexyl (2) + n-Butanol (3) and Corresponding Binaries at T = (293.15 to 333.15) K
rho1	878.84	kg/m3	303.15	Densities and Viscosities of Ternary System n-Dodecane (1) + Bicyclohexyl (2) + n-Butanol (3) and Corresponding Binaries at T = (293.15 to 333.15) K
rho1	875.25	kg/m3	308.15	Densities and Viscosities of Ternary System n-Dodecane (1) + Bicyclohexyl (2) + n-Butanol (3) and Corresponding Binaries at T = (293.15 to 333.15) K
rho1	871.66	kg/m3	313.15	Densities and Viscosities of Ternary System n-Dodecane (1) + Bicyclohexyl (2) + n-Butanol (3) and Corresponding Binaries at T = (293.15 to 333.15) K
rho1	868.08	kg/m3	318.15	Densities and Viscosities of Ternary System n-Dodecane (1) + Bicyclohexyl (2) + n-Butanol (3) and Corresponding Binaries at T = (293.15 to 333.15) K
rho1	864.50	kg/m3	323.15	Densities and Viscosities of Ternary System n-Dodecane (1) + Bicyclohexyl (2) + n-Butanol (3) and Corresponding Binaries at T = (293.15 to 333.15) K

rho1	860.93	kg/m3	328.15	Densities and Viscosities of Ternary System n-Dodecane (1) + Bicyclohexyl (2) + n-Butanol (3) and Corresponding Binaries at T = (293.15 to 333.15) K
rho1	850.29	kg/m3	343.15	Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K, Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and 1,2,3,4-Tetrahydronaphthalene or Trans-decahydronaphthalene at 0.1 MPa
rho1	828.60	kg/m3	373.15	Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K, Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and 1,2,3,4-Tetrahydronaphthalene or Trans-decahydronaphthalene at 0.1 MPa
rho1	857.35	kg/m3	333.15	Densities and Viscosities of Ternary System n-Dodecane (1) + Bicyclohexyl (2) + n-Butanol (3) and Corresponding Binaries at T = (293.15 to 333.15) K
sfust	25.89	J/molxK	273.50	NIST Webbook
sfust	24.46	J/molxK	277.20	NIST Webbook
sfust	6.01	J/molxK	256.10	NIST Webbook
sfust	2.77	J/molxK	267.50	NIST Webbook

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	372.50 ± 0.50	K	1.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.50496e+01
Coeff. B	-4.73244e+03
Coeff. C	-5.43230e+01
Temperature range (K), min.	374.91
Temperature range (K), max.	540.29

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	9.73446e+01
Coeff. B	-1.02483e+04
Coeff. C	-1.18827e+01
Coeff. D	5.45803e-06
Temperature range (K), min.	276.78
Temperature range (K), max.	727.00

## Sources

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C92513&Units=SI>

KDB Vapor Pressure Data:

<https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=591>

Density and Viscosity from 293.15 to 373.15 K, Speed of Sound and Bulk Modulus from 293.15 to 343.15 K,

<https://www.doi.org/10.1021/acs.jced.5b00790>

Surface Tension, and Flash Point of Binary Mixtures of Bicyclohexyl and

<https://www.thermo.com/files/research/kdb/mol/mol591.mol>

1,2,3,4-Tetrahydronaphthalene or Trans-decalin or naphthalene at 0.1 and Freezing Point for Binary Mixtures of the Yaws Handbook with Vapor Pressure

[https://www.chemedoc.com/doc/models/crippen\\_log10ws](https://www.chemedoc.com/doc/models/crippen_log10ws)

<https://www.doi.org/10.1021/je500275j>

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

of Bicyclohexane:



Densities and Viscosities of Ternary  
System n-Dodecane (1) + Bicyclohexyl  
(2) + Ethanol (3) and Corresponding  
Binaries at T = (293.15 to 333.15) K:  
Crippen Method:

<https://www.doi.org/10.1021/acs.jced.8b00438>

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rfi:</b>	Refractive Index
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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