

Adamantan-4-one, 1-methoxy

Inchi:	InChI=1S/C11H16O2/c1-13-11-4-7-2-8(5-11)10(12)9(3-7)6-11/h7-9H,2-6H2,1H3
InchiKey:	PKPPAWGCOMBLEA-UHFFFAOYSA-N
Formula:	C11H16O2
SMILES:	COC12CC3CC(C1)C(=O)C(C3)C2
Mol. weight [g/mol]:	180.24

Physical Properties

Property code	Value	Unit	Source
gf	-28.90	kJ/mol	Joback Method
hf	-333.15	kJ/mol	Joback Method
hfus	12.02	kJ/mol	Joback Method
hvap	45.19	kJ/mol	Joback Method
log10ws	-1.87		Crippen Method
logp	1.781		Crippen Method
mvol	140.710	ml/mol	McGowan Method
pc	2989.32	kPa	Joback Method
rinpol	1485.00		NIST Webbook
rinpol	1485.00		NIST Webbook
tb	561.38	K	Joback Method
tc	794.80	K	Joback Method
tf	374.14	K	Joback Method
vc	0.536	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	384.96	J/mol×K	561.38	Joback Method
cpg	404.36	J/mol×K	600.28	Joback Method
cpg	422.47	J/mol×K	639.19	Joback Method
cpg	439.44	J/mol×K	678.09	Joback Method
cpg	455.46	J/mol×K	716.99	Joback Method
cpg	470.71	J/mol×K	755.90	Joback Method
cpg	485.34	J/mol×K	794.80	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R44405&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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