

# Bicyclovetivenol

<b>Inchi:</b>	InChI=1S/C15H26O/c1-10-6-5-7-11(2)14-9-12(8-13(10)14)15(3,4)16/h11-14,16H,1,5-9H2
<b>InchiKey:</b>	VCECLBDDLTXHHP-NTXGFPLRSA-N
<b>Formula:</b>	C15H26O
<b>SMILES:</b>	C=C1CCCC(C)C2CC(C(C)(C)O)CC12
<b>Mol. weight [g/mol]:</b>	222.37

## Physical Properties

Property code	Value	Unit	Source
gf	52.20	kJ/mol	Joback Method
hf	-349.39	kJ/mol	Joback Method
hfus	20.13	kJ/mol	Joback Method
hvap	64.42	kJ/mol	Joback Method
log10ws	-4.15		Crippen Method
logp	3.776		Crippen Method
mcvol	202.060	ml/mol	McGowan Method
pc	2003.70	kPa	Joback Method
rinpol	1773.00		NIST Webbook
rinpol	1773.00		NIST Webbook
rinpol	1773.00		NIST Webbook
ripol	2604.00		NIST Webbook
ripol	2604.00		NIST Webbook
tb	651.93	K	Joback Method
tc	856.02	K	Joback Method
tf	349.05	K	Joback Method
vc	0.748	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	602.16	J/molxK	651.93	Joback Method
cpg	622.83	J/molxK	685.95	Joback Method
cpg	642.24	J/molxK	719.96	Joback Method
cpg	660.44	J/molxK	753.98	Joback Method
cpg	677.48	J/molxK	787.99	Joback Method

cpg	693.44	J/mol×K	822.01	Joback Method
cpg	708.35	J/mol×K	856.02	Joback Method
dvisc	0.0059008	Paxs	349.05	Joback Method
dvisc	0.0020979	Paxs	399.53	Joback Method
dvisc	0.0009406	Paxs	450.01	Joback Method
dvisc	0.0004958	Paxs	500.49	Joback Method
dvisc	0.0002939	Paxs	550.97	Joback Method
dvisc	0.0001902	Paxs	601.45	Joback Method
dvisc	0.0001317	Paxs	651.93	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R324451&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R324451&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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