

# Fenfluramine

**Other names:**

Benzeneethanamine, N-ethyl-«alpha»-methyl-3-(trifluoromethyl)-  
Phenethylamine, N-ethyl-«alpha»-methyl-m-(trifluoromethyl)-  
N-Ethyl-«alpha»-methyl-3-trifluoromethylphenethylamine  
3-(Trifluoromethyl)-N-ethyl-«alpha»-methylphenethylamine  
1-(meta-Trifluoromethyl-phenyl)-2 ethylaminopropane  
Acino  
Obedrex  
Pesos  
Ponderax PA  
Rotondin  
DL-Fenfluramine

**Inchi:**

InChI=1S/C12H16F3N/c1-3-16-9(2)7-10-5-4-6-11(8-10)12(13,14)15/h4-6,8-9,16H,3,7H2,

**InchiKey:**

DBGIVFWFUFKIQN-UHFFFAOYSA-N

**Formula:**

C12H16F3N

**SMILES:**

CCNC(C)Cc1cccc(C(F)(F)F)c1

**Mol. weight [g/mol]:**

231.26

**CAS:**

458-24-2

## Physical Properties

Property code	Value	Unit	Source
gf	-341.70	kJ/mol	Joback Method
hf	-614.84	kJ/mol	Joback Method
hfus	23.89	kJ/mol	Joback Method
hvap	47.55	kJ/mol	Joback Method
log10ws	-3.93		Crippen Method
logp	3.246		Crippen Method
mcvol	171.470	ml/mol	McGowan Method
pc	2165.35	kPa	Joback Method
rinpol	1226.00		NIST Webbook
rinpol	1240.60		NIST Webbook
rinpol	1220.00		NIST Webbook
rinpol	1220.00		NIST Webbook
rinpol	1220.00		NIST Webbook
rinpol	1173.00		NIST Webbook
rinpol	1177.00		NIST Webbook
rinpol	1185.00		NIST Webbook
rinpol	1220.00		NIST Webbook

rinpol	1183.00		NIST Webbook
rinpol	1222.00		NIST Webbook
rinpol	1220.00		NIST Webbook
rinpol	1240.60		NIST Webbook
rinpol	1183.00		NIST Webbook
rinpol	1240.00		NIST Webbook
rinpol	1220.00		NIST Webbook
rinpol	1220.00		NIST Webbook
ripol	1555.00		NIST Webbook
ripol	1562.00		NIST Webbook
ripol	1569.00		NIST Webbook
ripol	1531.00		NIST Webbook
ripol	1532.00		NIST Webbook
tb	549.93	K	Joback Method
tc	738.82	K	Joback Method
tf	305.79	K	Joback Method
vc	0.671	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	426.52	J/mol×K	549.93	Joback Method
cpg	441.97	J/mol×K	581.41	Joback Method
cpg	456.50	J/mol×K	612.89	Joback Method
cpg	470.15	J/mol×K	644.37	Joback Method
cpg	482.98	J/mol×K	675.86	Joback Method
cpg	495.02	J/mol×K	707.34	Joback Method
cpg	506.31	J/mol×K	738.82	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C458242&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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