

# 4-Ethyl-2-methylthiazoline

<b>Other names:</b>	2-Thiazoline, 4-ethyl-2-methyl-Thiazole, 4-ethyl-4,5-dihydro-2-methyl-4-Ethyl-2-methyl-2-thiazoline 2-Methyl-4-ethylthiazoline 4-Ethyl-2-methyl-delta <sup>2</sup> -thiazoline 4-ethyl-2-methyl-4,5-dihydrothiazole 4-Ethyl-2-methyl-delta
<b>Inchi:</b>	InChI=1S/C6H11NS/c1-3-6-4-8-5(2)7-6/h6H,3-4H2,1-2H3
<b>InchiKey:</b>	LENVRVUQOXYSDH-UHFFFAOYSA-N
<b>Formula:</b>	C6H11NS
<b>SMILES:</b>	CCC1CSC(C)=N1
<b>Mol. weight [g/mol]:</b>	129.22
<b>CAS:</b>	4293-61-2

## Physical Properties

Property code	Value	Unit	Source
gf	213.16	kJ/mol	Joback Method
hf	55.85	kJ/mol	Joback Method
hfus	14.86	kJ/mol	Joback Method
hvap	42.18	kJ/mol	Joback Method
log10ws	-1.88		Crippen Method
logp	1.930		Crippen Method
mvol	106.570	ml/mol	McGowan Method
pc	3916.03	kPa	Joback Method
rinpol	966.00		NIST Webbook
tb	457.63	K	Joback Method
tc	687.26	K	Joback Method
tf	336.55	K	Joback Method
vc	0.394	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	220.77	J/molxK	457.63	Joback Method

cpg	235.16	J/mol×K	495.90	Joback Method
cpg	248.82	J/mol×K	534.17	Joback Method
cpg	261.73	J/mol×K	572.45	Joback Method
cpg	273.93	J/mol×K	610.72	Joback Method
cpg	285.40	J/mol×K	648.99	Joback Method
cpg	296.17	J/mol×K	687.26	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4293612&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4293612&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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