

# Benzene, 1-bromo-2-ethyl-

<b>Other names:</b>	1-Bromo-2-ethylbenzene 1-Ethyl-2-bromobenzene 2-Bromo-1-ethylbenzene 2-Ethylbromobenzene o-Bromoethylbenzene
<b>Inchi:</b>	InChI=1S/C8H9Br/c1-2-7-5-3-4-6-8(7)9/h3-6H,2H2,1H3
<b>InchiKey:</b>	HVRUGFJYCAFAAN-UHFFFAOYSA-N
<b>Formula:</b>	C8H9Br
<b>SMILES:</b>	CCc1cccc1Br
<b>Mol. weight [g/mol]:</b>	185.06
<b>CAS:</b>	1973-22-4

## Physical Properties

Property code	Value	Unit	Source
gf	133.58	kJ/mol	Joback Method
hf	42.94	kJ/mol	Joback Method
hfus	15.41	kJ/mol	Joback Method
hvap	42.78	kJ/mol	Joback Method
log10ws	-3.44		Crippen Method
logp	3.012		Crippen Method
mcvol	117.320	ml/mol	McGowan Method
pc	3901.37	kPa	Joback Method
rinpol	1166.00		NIST Webbook
rinpol	1166.00		NIST Webbook
tb	472.50	K	NIST Webbook
tb	474.65 ± 0.07	K	NIST Webbook
tc	710.10	K	Joback Method
tf	205.61 ± 0.05	K	NIST Webbook
tf	205.90 ± 0.70	K	NIST Webbook
vc	0.438	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	272.56	J/molxK	710.10	Joback Method
cpg	264.29	J/molxK	671.79	Joback Method
cpg	255.40	J/molxK	633.48	Joback Method
cpg	245.87	J/molxK	595.18	Joback Method
cpg	235.65	J/molxK	556.87	Joback Method
cpg	224.69	J/molxK	518.57	Joback Method
cpg	212.97	J/molxK	480.26	Joback Method
dvisc	0.0021418	Paxs	278.66	Joback Method
dvisc	0.0002886	Paxs	480.26	Joback Method
dvisc	0.0003554	Paxs	446.66	Joback Method
dvisc	0.0004529	Paxs	413.06	Joback Method
dvisc	0.0006024	Paxs	379.46	Joback Method
dvisc	0.0008469	Paxs	345.86	Joback Method
dvisc	0.0012813	Paxs	312.26	Joback Method
hvapt	48.10	kJ/mol	445.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.40866e+01
Coeff. B	-3.78741e+03
Coeff. C	-7.24910e+01
Temperature range (K), min.	346.96
Temperature range (K), max.	504.10

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1973224&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1973224&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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