

# Pentorex

<b>Inchi:</b>	InChI=1S/C11H17N/c1-9(11(2,3)12)10-7-5-4-6-8-10/h4-9H,12H2,1-3H3
<b>InchiKey:</b>	UMWAUEZOGHNSCH-UHFFFAOYSA-N
<b>Formula:</b>	C11H17N
<b>SMILES:</b>	CC(c1ccccc1)C(C)(C)N
<b>Mol. weight [g/mol]:</b>	163.26
<b>CAS:</b>	434-43-5

## Physical Properties

Property code	Value	Unit	Source
gf	221.00	kJ/mol	Joback Method
hf	-14.08	kJ/mol	Joback Method
hfus	12.55	kJ/mol	Joback Method
hvap	51.31	kJ/mol	Joback Method
log10ws	-3.04		Crippen Method
logp	2.527		Crippen Method
mcvol	152.070	ml/mol	McGowan Method
pc	2893.62	kPa	Joback Method
rinpol	1261.00		NIST Webbook
rinpol	1256.00		NIST Webbook
rinpol	1256.00		NIST Webbook
rinpol	1256.00		NIST Webbook
rinpol	1261.00		NIST Webbook
ripol	1648.00		NIST Webbook
ripol	1648.00		NIST Webbook
ripol	1648.00		NIST Webbook
tb	546.62	K	Joback Method
tc	777.66	K	Joback Method
tf	310.83	K	Joback Method
vc	0.555	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	366.25	J/mol×K	546.62	Joback Method

cpg	383.56	J/mol×K	585.13	Joback Method
cpg	399.61	J/mol×K	623.63	Joback Method
cpg	414.48	J/mol×K	662.14	Joback Method
cpg	428.24	J/mol×K	700.65	Joback Method
cpg	440.98	J/mol×K	739.16	Joback Method
cpg	452.77	J/mol×K	777.66	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C434435&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C434435&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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