

# 3-Tetradecanone

<b>Other names:</b>	tetradecan-3-one
<b>Inchi:</b>	InChI=1S/C14H28O/c1-3-5-6-7-8-9-10-11-12-13-14(15)4-2/h3-13H2,1-2H3
<b>InchiKey:</b>	OCHYRSKMMMYUMI-UHFFFAOYSA-N
<b>Formula:</b>	C14H28O
<b>SMILES:</b>	CCCCCCCCCCCC(=O)CC
<b>Mol. weight [g/mol]:</b>	212.37
<b>CAS:</b>	629-23-2

## Physical Properties

Property code	Value	Unit	Source
gf	-61.92	kJ/mol	Joback Method
hf	-444.87	kJ/mol	Joback Method
hfus	33.61	kJ/mol	Joback Method
hvap	53.50	kJ/mol	Joback Method
log10ws	-4.96		Crippen Method
logp	4.886		Crippen Method
mcvol	209.690	ml/mol	McGowan Method
pc	1601.28	kPa	Joback Method
rhoc	237.86 ± 6.37	kg/m3	NIST Webbook
rinpol	1568.00		NIST Webbook
rinpol	1568.00		NIST Webbook
rinpol	1585.00		NIST Webbook
rinpol	1585.00		NIST Webbook
rinpol	1573.00		NIST Webbook
ripol	1855.00		NIST Webbook
ripol	1855.00		NIST Webbook
tb	573.59	K	Joback Method
tc	726.80 ± 3.70	K	NIST Webbook
tf	297.47	K	Joback Method
vc	0.826	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	537.57	J/molxK	573.59	Joback Method
cpg	554.73	J/molxK	601.60	Joback Method
cpg	571.17	J/molxK	629.62	Joback Method
cpg	586.91	J/molxK	657.63	Joback Method
cpg	601.98	J/molxK	685.64	Joback Method
cpg	616.40	J/molxK	713.66	Joback Method
cpg	630.18	J/molxK	741.67	Joback Method
dvisc	0.0040142	Paxs	297.47	Joback Method
dvisc	0.0017253	Paxs	343.49	Joback Method
dvisc	0.0009053	Paxs	389.51	Joback Method
dvisc	0.0005444	Paxs	435.53	Joback Method
dvisc	0.0003608	Paxs	481.55	Joback Method
dvisc	0.0002569	Paxs	527.57	Joback Method
dvisc	0.0001932	Paxs	573.59	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	425.20	K	2.10	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.53929e+01
Coeff. B	-4.91033e+03
Coeff. C	-9.46480e+01
Temperature range (K), min.	419.72
Temperature range (K), max.	581.71

## Sources

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C629232&Units=SI>

**The Yaws Handbook of Vapor**

**Pressure:**

**Crippen Method:**

**Crippen Method:**

**Joback Method:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rhoc:</b>	Critical density
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/68-951-1/3-Tetradecanone.pdf>

Generated by Cheméo on 2024-04-25 20:41:45.877812532 +0000 UTC m=+16366954.798389848.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.