

# Prezizaene

<b>Other names:</b>	Preziza-7(15)-ene
<b>Inchi:</b>	InChI=1S/C15H24/c1-10-5-6-13-14(3,4)11(2)12-7-8-15(10,13)9-12/h10,12-13H,2,5-9H2,7-8H
<b>InchiKey:</b>	RFSYBMDOYOBTCL-RZLPUIOOSA-N
<b>Formula:</b>	C15H24
<b>SMILES:</b>	<chem>C=C1C2CCC3(C2)C(C)CCC3C1(C)C</chem>
<b>Mol. weight [g/mol]:</b>	204.35
<b>CAS:</b>	116498-31-8

## Physical Properties

Property code	Value	Unit	Source
gf	260.15	kJ/mol	Joback Method
hf	-72.81	kJ/mol	Joback Method
hfus	13.20	kJ/mol	Joback Method
hvap	46.31	kJ/mol	Joback Method
log10ws	-4.43		Crippen Method
logp	4.415		Crippen Method
mcvol	185.330	ml/mol	McGowan Method
pc	2113.89	kPa	Joback Method
rinpol	1449.00		NIST Webbook
rinpol	1450.00		NIST Webbook
rinpol	1432.00		NIST Webbook
rinpol	1450.00		NIST Webbook
rinpol	1452.00		NIST Webbook
rinpol	1448.00		NIST Webbook
ripol	1632.00		NIST Webbook
ripol	1632.00		NIST Webbook
ripol	1590.00		NIST Webbook
tb	561.66	K	Joback Method
tc	785.91	K	Joback Method
tf	358.59	K	Joback Method
vc	0.709	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	501.94	J/mol×K	561.66	Joback Method
cpg	525.20	J/mol×K	599.03	Joback Method
cpg	546.85	J/mol×K	636.41	Joback Method
cpg	567.18	J/mol×K	673.78	Joback Method
cpg	586.48	J/mol×K	711.16	Joback Method
cpg	605.02	J/mol×K	748.53	Joback Method
cpg	623.10	J/mol×K	785.91	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C116498318&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C116498318&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/68-982-7/Prezizaene.pdf>

Generated by Cheméo on 2024-04-29 22:38:03.190055001 +0000 UTC m=+16719532.110632325.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.