

# Benzamide, N-(4-chloro-9,10-dihydro-9,10-dioxo-1-anthraceny

Other names:

Benzamide, N-(4-chloro-1-anthraquinonyl)-

Anthraquinone, 1-benzamido-4-chloro-

Chlorobenzene

1-Benzamido-4-chloroanthraquinone

1-Benzoylamino-4-chloroanthraquinone

1-Chloro-4-benzamidoanthraquinone

1-Chloro-4-benzoylaminoanthraquinone

1-(Benzoylamino)-2-chloroanthraquinone

1Ba-4-Xa

4-Benzamido-1-chloroanthraquinone

NSC 39954

N-(4-chloro-9,10-dihydro-9,10-dioxo-1-anthryl)benzamide

**Inchi:** InChI=1S/C21H12ClNO3/c22-15-10-11-16(23-21(26)12-6-2-1-3-7-12)18-17(15)19(24)13

**InchiKey:** FNCVZYRPXOZNSM-UHFFFAOYSA-N

**Formula:** C21H12ClNO3

**SMILES:** O=C(Nc1ccc(Cl)c2c1C(=O)c1cccc1C2=O)c1cccc1

**Mol. weight [g/mol]:** 361.78

**CAS:** 81-45-8

## Physical Properties

Property code	Value	Unit	Source
gf	208.57	kJ/mol	Joback Method
hf	-64.01	kJ/mol	Joback Method
hfus	39.79	kJ/mol	Joback Method
hvap	97.93	kJ/mol	Joback Method
log10ws	-6.29		Crippen Method
logp	4.368		Crippen Method
mcvol	251.540	ml/mol	McGowan Method
pc	2370.28	kPa	Joback Method
tb	1064.09	K	Joback Method
tc	1343.49	K	Joback Method
tf	750.42	K	Joback Method
vc	0.958	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	753.03	J/mol×K	1064.09	Joback Method
cpg	761.73	J/mol×K	1110.66	Joback Method
cpg	769.10	J/mol×K	1157.22	Joback Method
cpg	775.26	J/mol×K	1203.79	Joback Method
cpg	780.29	J/mol×K	1250.36	Joback Method
cpg	784.29	J/mol×K	1296.93	Joback Method
cpg	787.35	J/mol×K	1343.49	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C81458&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C81458&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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