

# (CH<sub>3</sub>)<sub>2</sub>N-CH=N-phenyl

**Inchi:** InChI=1S/C9H12N2/c1-11(2)8-10-9-6-4-3-5-7-9/h3-8H,1-2H3/b10-8+  
**InchiKey:** SRPCLECGIYMIMN-CSKARUKUSA-N  
**Formula:** C<sub>9</sub>H<sub>12</sub>N<sub>2</sub>  
**SMILES:** CN(C)C=Nc1ccccc1  
**Mol. weight [g/mol]:** 148.21  
**CAS:** 56687-95-7

## Physical Properties

Property code	Value	Unit	Source
affp	983.80	kJ/mol	NIST Webbook
basg	951.30	kJ/mol	NIST Webbook
hf	157.19	kJ/mol	Joback Method
hvap	43.26	kJ/mol	Joback Method
log10ws	-1.52		Crippen Method
logp	1.908		Crippen Method
mcvol	129.570	ml/mol	McGowan Method
pc	2896.73	kPa	Joback Method
tb	521.12	K	Joback Method
tc	745.87	K	Joback Method

## Sources

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C56687957&Units=SI>  
**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307I>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)  
**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

**affp:** Proton affinity

<b>basg:</b>	Gas basicity
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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