

2-Naphthalenecarboxylic acid, methyl ester

Other names:	2-Naphthoic acid, methyl ester Methyl 2-naphthalenecarboxylate Methyl 2-naphthoate
Inchi:	InChI=1S/C12H10O2/c1-14-12(13)11-7-6-9-4-2-3-5-10(9)8-11/h2-8H,1H3
InchiKey:	IODOXLFXNATGI-UHFFFAOYSA-N
Formula:	C12H10O2
SMILES:	<chem>COC(=O)c1ccc2ccccc2c1</chem>
Mol. weight [g/mol]:	186.21
CAS:	2459-25-8

Physical Properties

Property code	Value	Unit	Source
gf	25.67	kJ/mol	Joback Method
hf	-119.68	kJ/mol	Joback Method
hfus	20.29	kJ/mol	Joback Method
hvap	56.04	kJ/mol	Joback Method
log10ws	-3.52		Crippen Method
logp	2.626		Crippen Method
mcvol	144.160	ml/mol	McGowan Method
pc	3217.33	kPa	Joback Method
rinpol	1651.00		NIST Webbook
rinpol	1646.00		NIST Webbook
rinpol	279.47		NIST Webbook
rinpol	279.47		NIST Webbook
tb	600.89	K	Joback Method
tc	836.14	K	Joback Method
tf	350.00 ± 0.10	K	NIST Webbook
vc	0.545	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	402.88	J/molxK	836.14	Joback Method
cpg	393.88	J/molxK	796.93	Joback Method

cpg	384.14	J/mol×K	757.72	Joback Method
cpg	373.60	J/mol×K	718.51	Joback Method
cpg	362.21	J/mol×K	679.31	Joback Method
cpg	349.90	J/mol×K	640.10	Joback Method
cpg	336.63	J/mol×K	600.89	Joback Method
dvisc	0.0014170	Paxs	368.80	Joback Method
dvisc	0.0002966	Paxs	600.89	Joback Method
dvisc	0.0003519	Paxs	562.21	Joback Method
dvisc	0.0004282	Paxs	523.53	Joback Method
dvisc	0.0005377	Paxs	484.85	Joback Method
dvisc	0.0007022	Paxs	446.16	Joback Method
dvisc	0.0009648	Paxs	407.48	Joback Method
hfust	27.10	kJ/mol	350.20	NIST Webbook

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C2459258&Units=SI

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature

tf: Normal melting (fusion) point

vc: Critical Volume

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