

# Propionic acid, 2,2-dichloro-, pentyl ester

<b>Other names:</b>	Propanoic acid, 2,2-dichloro-, pentyl ester
<b>Inchi:</b>	InChI=1S/C8H14Cl2O2/c1-3-4-5-6-12-7(11)8(2,9)10/h3-6H2,1-2H3
<b>InchiKey:</b>	ZCPOSFLAFHCNHBE-UHFFFAOYSA-N
<b>Formula:</b>	C8H14Cl2O2
<b>SMILES:</b>	CCCCCOC(=O)C(C)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	213.10
<b>CAS:</b>	17640-08-3

## Physical Properties

Property code	Value	Unit	Source
gf	-238.46	kJ/mol	Joback Method
hf	-493.48	kJ/mol	Joback Method
hfus	20.24	kJ/mol	Joback Method
hvap	50.03	kJ/mol	Joback Method
log10ws	-2.95		Crippen Method
logp	2.914		Crippen Method
mcvol	155.500	ml/mol	McGowan Method
pc	2512.55	kPa	Joback Method
tb	530.36	K	Joback Method
tc	727.45	K	Joback Method
tf	314.34	K	Joback Method
vc	0.595	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	337.84	J/mol×K	530.36	Joback Method
cpg	350.00	J/mol×K	563.21	Joback Method
cpg	361.50	J/mol×K	596.06	Joback Method
cpg	372.35	J/mol×K	628.90	Joback Method
cpg	382.59	J/mol×K	661.75	Joback Method
cpg	392.23	J/mol×K	694.60	Joback Method
cpg	401.29	J/mol×K	727.45	Joback Method
dvisc	0.0032784	Paxs	314.34	Joback Method

dvisc	0.0016965	Paxs	350.34	Joback Method
dvisc	0.0009926	Paxs	386.35	Joback Method
dvisc	0.0006363	Paxs	422.35	Joback Method
dvisc	0.0004374	Paxs	458.35	Joback Method
dvisc	0.0003176	Paxs	494.36	Joback Method
dvisc	0.0002408	Paxs	530.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17640083&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17640083&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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