

1-Nitropyrene

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|-----------------------------|--|
| Other names: | 3-Nitropyrene Pyrene, 1-nitro- |
| Inchi: | InChI=1S/C16H9NO2/c18-17(19)14-9-7-12-5-4-10-2-1-3-11-6-8-13(14)16(12)15(10)11/h |
| InchiKey: | ALRLPDGCPYIVHP-UHFFFAOYSA-N |
| Formula: | C16H9NO2 |
| SMILES: | O=[N+]([O-])c1ccc2ccc3cccc4ccc1c2c34 |
| Mol. weight [g/mol]: | 247.25 |
| CAS: | 5522-43-0 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | 517.10 | kJ/mol | Joback Method |
| hf | 345.54 | kJ/mol | Joback Method |
| hfus | 35.47 | kJ/mol | Joback Method |
| hvap | 76.35 | kJ/mol | Joback Method |
| log10ws | -6.95 | | Crippen Method |
| logp | 4.492 | | Crippen Method |
| mcvol | 175.880 | ml/mol | McGowan Method |
| pc | 3062.55 | kPa | Joback Method |
| rinpol | 421.08 | | NIST Webbook |
| rinpol | 425.50 | | NIST Webbook |
| rinpol | 421.53 | | NIST Webbook |
| rinpol | 420.60 | | NIST Webbook |
| rinpol | 421.48 | | NIST Webbook |
| rinpol | 419.25 | | NIST Webbook |
| rinpol | 421.13 | | NIST Webbook |
| rinpol | 421.18 | | NIST Webbook |
| rinpol | 421.39 | | NIST Webbook |
| rinpol | 421.48 | | NIST Webbook |
| rinpol | 419.25 | | NIST Webbook |
| rinpol | 421.08 | | NIST Webbook |
| rinpol | 421.26 | | NIST Webbook |
| rinpol | 421.27 | | NIST Webbook |
| tb | 808.18 | K | Joback Method |
| tc | 1082.43 | K | Joback Method |
| tf | 582.05 | K | Joback Method |
| vc | 0.702 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|---------------|---------|-----------------|---------------|
| cpg | 516.66 | J/mol×K | 991.01 | Joback Method |
| cpg | 527.49 | J/mol×K | 1036.72 | Joback Method |
| cpg | 473.39 | J/mol×K | 808.18 | Joback Method |
| cpg | 484.71 | J/mol×K | 853.89 | Joback Method |
| cpg | 495.53 | J/mol×K | 899.60 | Joback Method |
| cpg | 506.09 | J/mol×K | 945.31 | Joback Method |
| cpg | 538.85 | J/mol×K | 1082.43 | Joback Method |
| hfust | 18.90 | kJ/mol | 425.90 | NIST Webbook |
| hsubt | 125.20 ± 3.80 | kJ/mol | 393.50 | NIST Webbook |

Sources

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C5522430&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hfust: | Enthalpy of fusion at a given temperature |
| hsubt: | Enthalpy of sublimation at a given temperature |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |

| | |
|----------------|----------------------------------|
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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