

4H-Cyclopenta[def]phenanthrenequinone

Inchi:	InChI=1S/C15H8O2/c16-14-10-5-1-3-8-7-9-4-2-6-11(15(14)17)13(9)12(8)10/h1-6H,7H2
InchiKey:	AHDICIQOAZDBEN-UHFFFAOYSA-N
Formula:	C15H8O2
SMILES:	O=C1C(=O)c2cccc3c2-c2c(cccc21)C3
Mol. weight [g/mol]:	220.22

Physical Properties

Property code	Value	Unit	Source
gf	189.76	kJ/mol	Joback Method
hf	3.61	kJ/mol	Joback Method
hfus	20.58	kJ/mol	Joback Method
hvap	64.61	kJ/mol	Joback Method
log10ws	-4.80		Crippen Method
logp	2.637		Crippen Method
mvol	156.110	ml/mol	McGowan Method
pc	3322.01	kPa	Joback Method
rinpol	342.64		NIST Webbook
rinpol	2195.00		NIST Webbook
tb	761.53	K	Joback Method
tc	1031.86	K	Joback Method
tf	553.09	K	Joback Method
vc	0.614	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	424.33	J/mol×K	761.53	Joback Method
cpg	436.98	J/mol×K	806.59	Joback Method
cpg	448.77	J/mol×K	851.64	Joback Method
cpg	459.82	J/mol×K	896.70	Joback Method
cpg	470.26	J/mol×K	941.75	Joback Method
cpg	480.22	J/mol×K	986.81	Joback Method
cpg	489.83	J/mol×K	1031.86	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R553109&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rlnol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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