

# Ethyl 2,4,5-trichlorophenyl carbonate

<b>Inchi:</b>	InChI=1S/C9H7Cl3O3/c1-2-14-9(13)15-8-4-6(11)5(10)3-7(8)12/h3-4H,2H2,1H3
<b>InchiKey:</b>	GPVCBFIVLNSTRS-UHFFFAOYSA-N
<b>Formula:</b>	C9H7Cl3O3
<b>SMILES:</b>	CCOC(=O)Oc1cc(Cl)c(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	269.51
<b>CAS:</b>	22876-29-5

## Physical Properties

Property code	Value	Unit	Source
gf	-266.29	kJ/mol	Joback Method
hf	-451.21	kJ/mol	Joback Method
hfus	28.51	kJ/mol	Joback Method
hvap	64.61	kJ/mol	Joback Method
log10ws	-4.32		Crippen Method
logp	4.182		Crippen Method
mcvol	163.940	ml/mol	McGowan Method
pc	2835.36	kPa	Joback Method
tb	657.94	K	Joback Method
tc	886.25	K	Joback Method
tf	439.32	K	Joback Method
vc	0.621	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	345.09	J/molxK	657.94	Joback Method
cpg	354.53	J/molxK	695.99	Joback Method
cpg	363.36	J/molxK	734.04	Joback Method
cpg	371.56	J/molxK	772.10	Joback Method
cpg	379.13	J/molxK	810.15	Joback Method
cpg	386.04	J/molxK	848.20	Joback Method
cpg	392.29	J/molxK	886.25	Joback Method
dvisc	0.0007498	Paxs	439.32	Joback Method
dvisc	0.0005253	Paxs	475.76	Joback Method

dvisc	0.0003871	Paxs	512.19	Joback Method
dvisc	0.0002970	Paxs	548.63	Joback Method
dvisc	0.0002356	Paxs	585.07	Joback Method
dvisc	0.0001920	Paxs	621.50	Joback Method
dvisc	0.0001601	Paxs	657.94	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C22876295&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C22876295&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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