

# 5-Aminoindan

<b>Other names:</b>	5-Aminohydrindene 5-Amino indane 1H-Inden-5-amine, 2,3-dihydro- indan-5-amine
<b>Inchi:</b>	InChI=1S/C9H11N/c10-9-5-4-7-2-1-3-8(7)6-9/h4-6H,1-3,10H2
<b>InchiKey:</b>	LEWZOBYWGWNCK-UHFFFAOYSA-N
<b>Formula:</b>	C9H11N
<b>SMILES:</b>	<chem>Nc1ccc2c(c1)CCC2</chem>
<b>Mol. weight [g/mol]:</b>	133.19
<b>CAS:</b>	24425-40-9

## Physical Properties

Property code	Value	Unit	Source
gf	252.96	kJ/mol	Joback Method
hf	111.43	kJ/mol	Joback Method
hfus	14.59	kJ/mol	Joback Method
hvap	50.09	kJ/mol	Joback Method
log10ws	-2.19		Crippen Method
logp	1.758		Crippen Method
mcvol	113.030	ml/mol	McGowan Method
pc	4103.88	kPa	Joback Method
rinpol	232.12		NIST Webbook
rinpol	231.35		NIST Webbook
tb	525.90	K	Joback Method
tc	765.83	K	Joback Method
tf	310.50 ± 0.50	K	NIST Webbook
vc	0.418	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	255.08	J/mol×K	525.90	Joback Method
cpg	268.83	J/mol×K	565.89	Joback Method
cpg	281.53	J/mol×K	605.88	Joback Method

cpg	293.27	J/mol×K	645.87	Joback Method
cpg	304.13	J/mol×K	685.86	Joback Method
cpg	314.19	J/mol×K	725.85	Joback Method
cpg	323.54	J/mol×K	765.83	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	521.20	K	99.30	NIST Webbook
tbrp	521.00 ± 1.00	K	99.30	NIST Webbook
tbrp	419.50 ± 0.50	K	3.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24425409&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24425409&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/69-176-1/5-Aminoindan.pdf>

Generated by Cheméo on 2024-04-26 09:27:09.753727013 +0000 UTC m=+16412878.674304328.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.