

# Propanedioic acid, dimethyl ester

<b>Other names:</b>	1,3-propanedioic acid, dimethyl ester Dimethyl ester of malonic acid Dimethyl malonate Dimethyl propanedioate Malonic acid, dimethyl ester Methyl malonate dimethyl 1,3-propanedioate
<b>Inchi:</b>	InChI=1S/C5H8O4/c1-8-4(6)3-5(7)9-2/h3H2,1-2H3
<b>InchiKey:</b>	BEPAFCGSDWSTEL-UHFFFAOYSA-N
<b>Formula:</b>	C5H8O4
<b>SMILES:</b>	COC(=O)CC(=O)OC
<b>Mol. weight [g/mol]:</b>	132.11
<b>CAS:</b>	108-59-8

## Physical Properties

Property code	Value	Unit	Source
chl	-2311.20 ± 0.63	kJ/mol	NIST Webbook
chl	-2311.20 ± 0.63	kJ/mol	NIST Webbook
chs	-2318.00	kJ/mol	NIST Webbook
gf	-476.62	kJ/mol	Joback Method
hf	-737.80 ± 1.00	kJ/mol	NIST Webbook
hf	-737.80 ± 1.00	kJ/mol	NIST Webbook
hfl	-799.69 ± 0.63	kJ/mol	NIST Webbook
hfl	-799.69 ± 0.63	kJ/mol	NIST Webbook
hfus	14.28	kJ/mol	Joback Method
hsub	111.70 ± 2.10	kJ/mol	NIST Webbook
hvap	57.50 ± 0.30	kJ/mol	NIST Webbook
hvap	61.90	kJ/mol	NIST Webbook
hvap	61.84 ± 0.79	kJ/mol	NIST Webbook
hvap	61.84 ± 0.79	kJ/mol	NIST Webbook
log10ws	0.36		Crippen Method
logp	-0.278		Crippen Method
mcvol	96.190	ml/mol	McGowan Method

pc

3530.00

kPa

Vapor Pressures,  
Enthalpies of Vaporization,  
and Critical Parameters of  
a Series of Linear Aliphatic  
Dimethyl Esters of  
Dicarboxylic Acids

rinpol	917.00		NIST Webbook
rinpol	914.00		NIST Webbook
rinpol	917.00		NIST Webbook
rinpol	918.00		NIST Webbook
rinpol	891.00		NIST Webbook
rinpol	897.00		NIST Webbook
rinpol	899.00		NIST Webbook
rinpol	896.00		NIST Webbook
rinpol	888.00		NIST Webbook
rinpol	902.00		NIST Webbook
rinpol	895.00		NIST Webbook
rinpol	882.00		NIST Webbook
rinpol	910.00		NIST Webbook
rinpol	887.00		NIST Webbook
rinpol	914.00		NIST Webbook
rinpol	884.00		NIST Webbook
rinpol	895.00		NIST Webbook
rinpol	887.00		NIST Webbook
rinpol	897.00		NIST Webbook
rinpol	913.00		NIST Webbook
rinpol	908.00		NIST Webbook
rinpol	921.00		NIST Webbook
rinpol	910.00		NIST Webbook
rinpol	902.00		NIST Webbook
rinpol	915.00		NIST Webbook
ripol	1506.00		NIST Webbook
ripol	1472.00		NIST Webbook
ripol	1523.00		NIST Webbook
ripol	1489.00		NIST Webbook
ripol	1482.00		NIST Webbook
ripol	1484.00		NIST Webbook
ripol	1499.00		NIST Webbook
ripol	1489.00		NIST Webbook
ripol	1500.00		NIST Webbook
ripol	1482.00		NIST Webbook
tb	454.60 ± 0.80	K	NIST Webbook
tb	454.60 ± 0.50	K	NIST Webbook
tb	454.60	K	NIST Webbook
tc	656.26	K	Joback Method
tf	211.30 ± 1.00	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	196.70	J/molxK	466.38	Joback Method
cpg	204.67	J/molxK	498.03	Joback Method
cpg	212.42	J/molxK	529.67	Joback Method
cpg	219.95	J/molxK	561.32	Joback Method
cpg	227.23	J/molxK	592.97	Joback Method
cpg	234.25	J/molxK	624.62	Joback Method
cpg	240.99	J/molxK	656.26	Joback Method
cpl	210.10	J/molxK	298.15	NIST Webbook
dvisc	0.0002822	Paxs	466.38	Joback Method
dvisc	0.0012551	Paxs	319.75	Joback Method
dvisc	0.0020268	Paxs	290.43	Joback Method
dvisc	0.0006014	Paxs	378.40	Joback Method
dvisc	0.0004507	Paxs	407.73	Joback Method
dvisc	0.0003511	Paxs	437.06	Joback Method
dvisc	0.0008423	Paxs	349.08	Joback Method
hvapt	52.90 ± 0.20	kJ/mol	405.50	NIST Webbook
hvapt	49.50 ± 0.20	kJ/mol	405.50	NIST Webbook
hvapt	46.10 ± 0.30	kJ/mol	405.50	NIST Webbook
hvapt	61.80 ± 0.80	kJ/mol	293.00	NIST Webbook
hvapt	50.00	kJ/mol	497.00	NIST Webbook
hvapt	53.70	kJ/mol	381.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.69518e+01
Coeff. B	-5.19477e+03
Coeff. C	-3.34050e+01
Temperature range (K), min.	345.14
Temperature range (K), max.	479.68

# Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Vapor Pressures, Enthalpies of Vaporization, and Critical Parameters of a Series of Linear Aliphatic Dimethyl Esters of Dicarboxylic Acids:</b>	<a href="https://www.doi.org/10.1021/je0602418">https://www.doi.org/10.1021/je0602418</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C108598&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C108598&amp;Units=SI</a>

# Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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